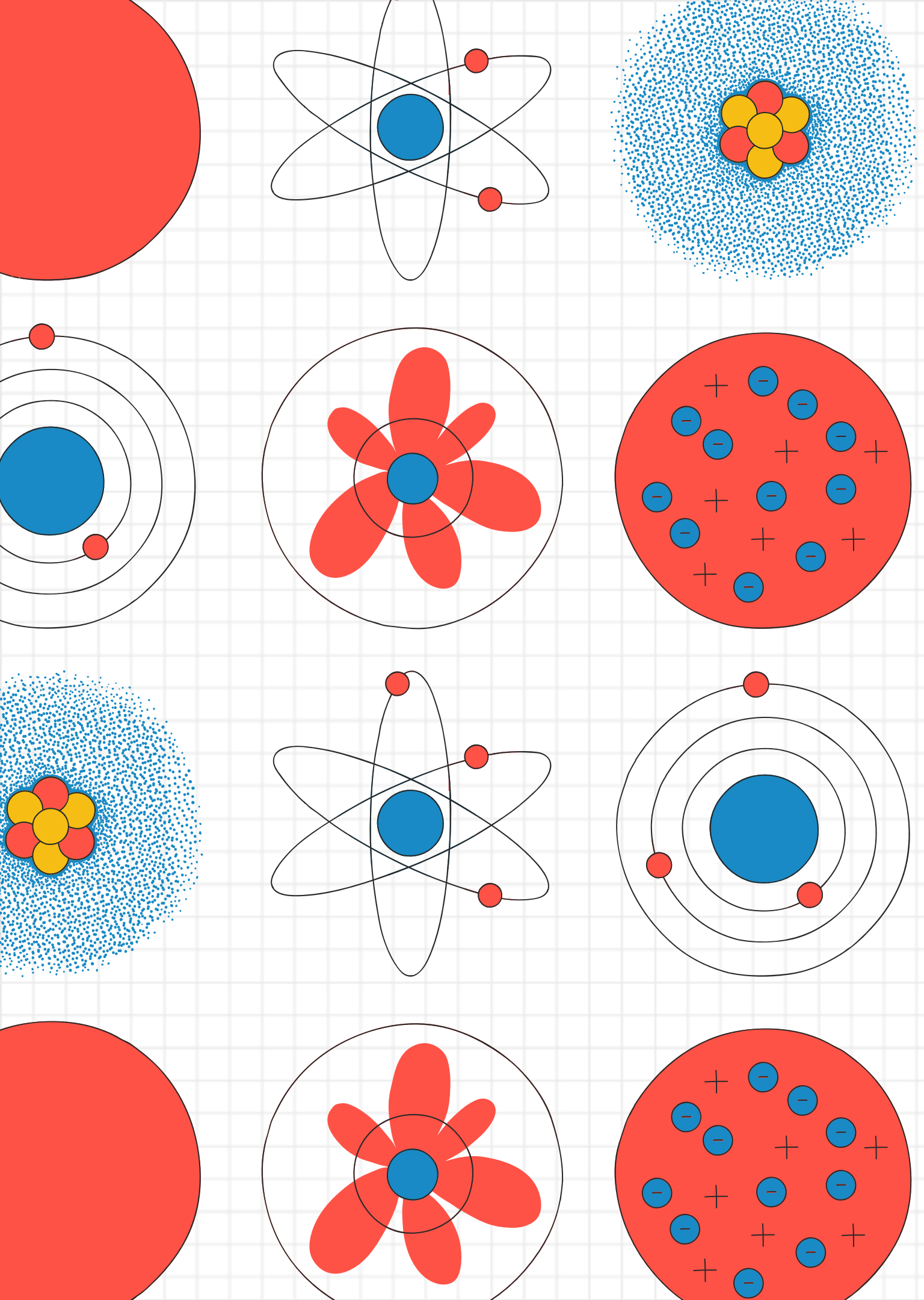




Nomenclatura de Compostos Orgânicos

Prof. Pedro



Na aula de Hoje

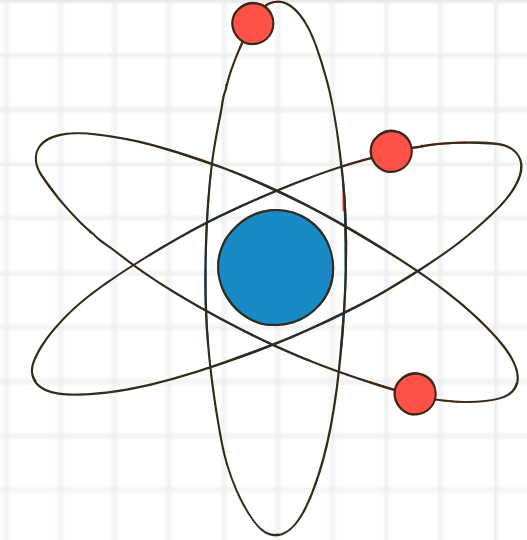
Revisão

Como surgiu a nomenclatura

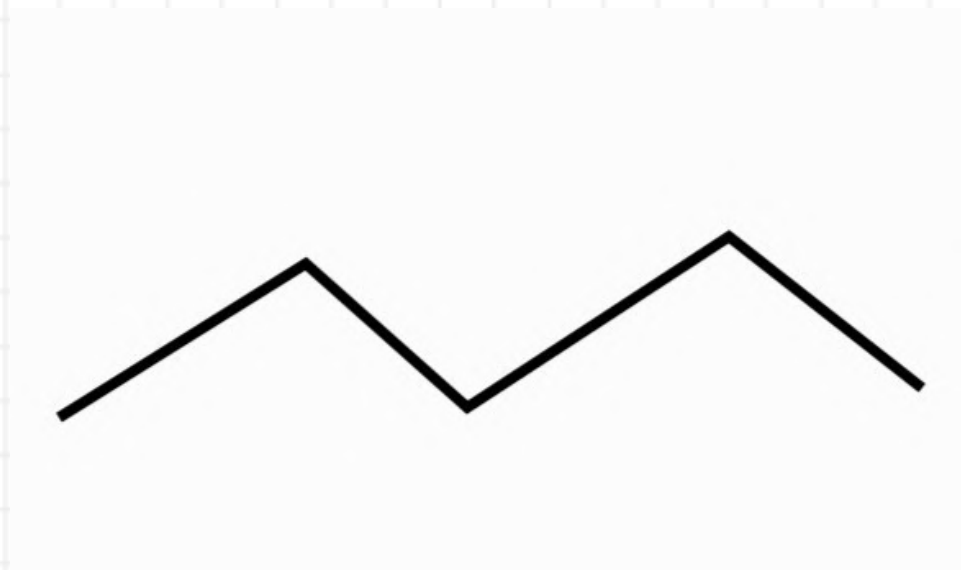
Regras para a nomenclatura

Exercícios

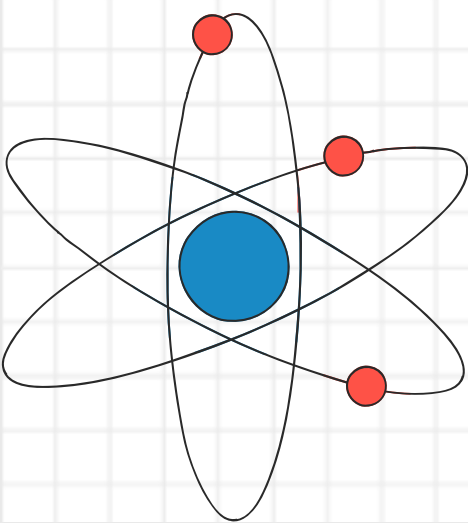
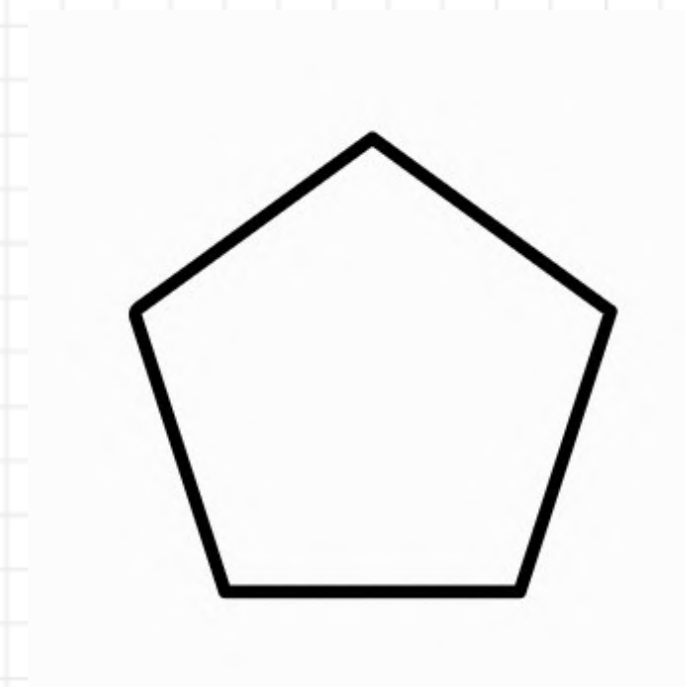
Revisão



Cadeia Aberta

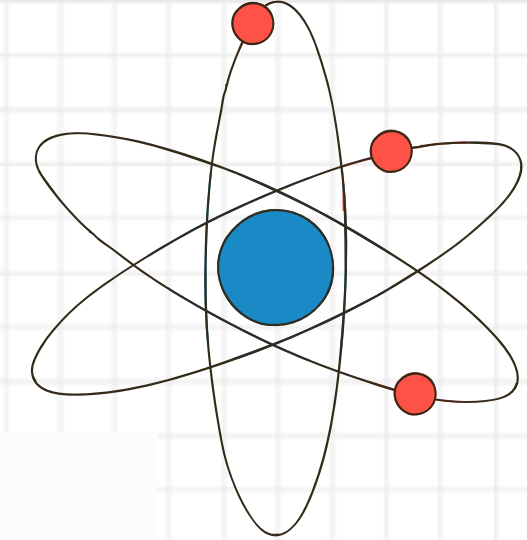


Cadeia Fechada



Revisão

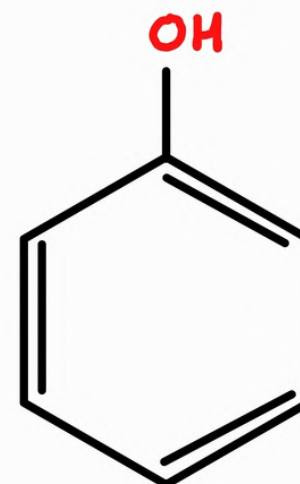
Grupos Funcionais



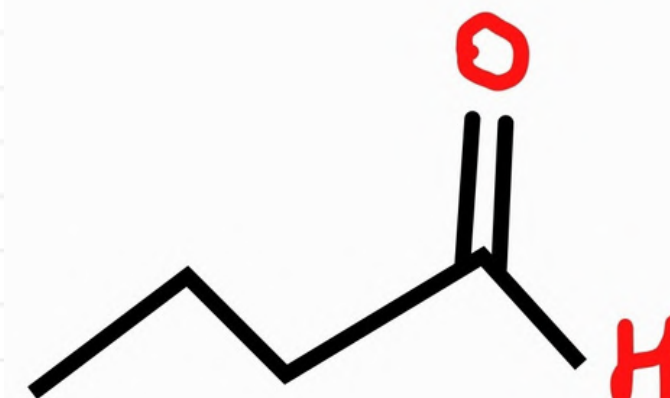
HIDROCARBONETOS



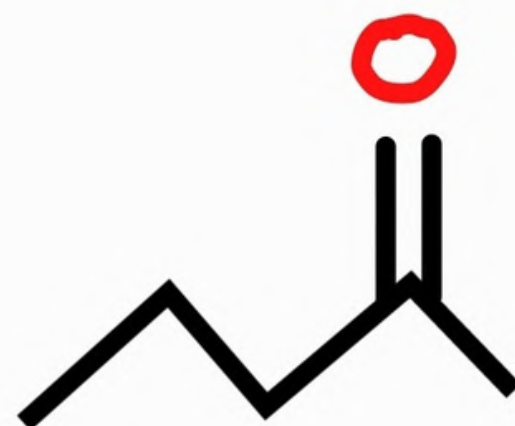
ÁLCOOL



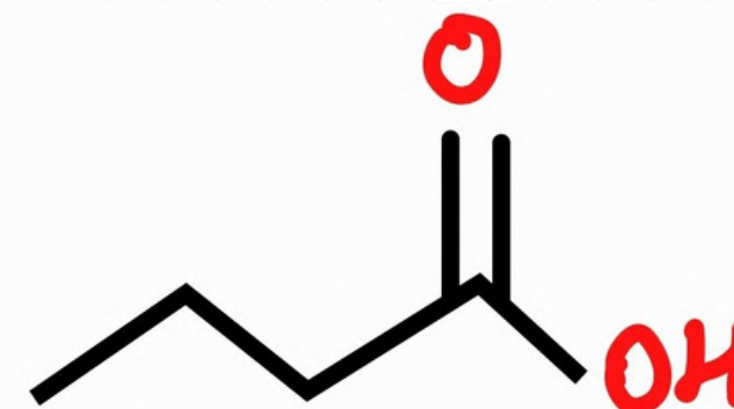
FENOL



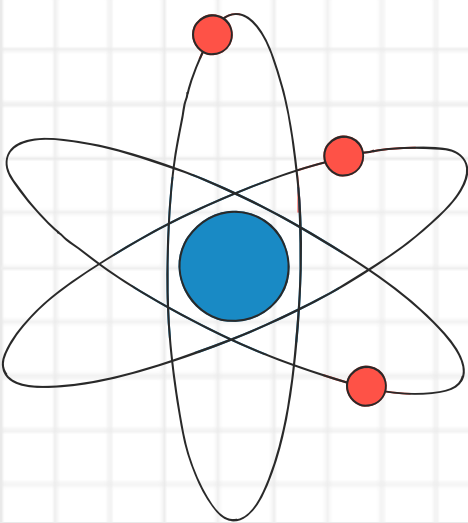
ALDEÍDO



CETONA

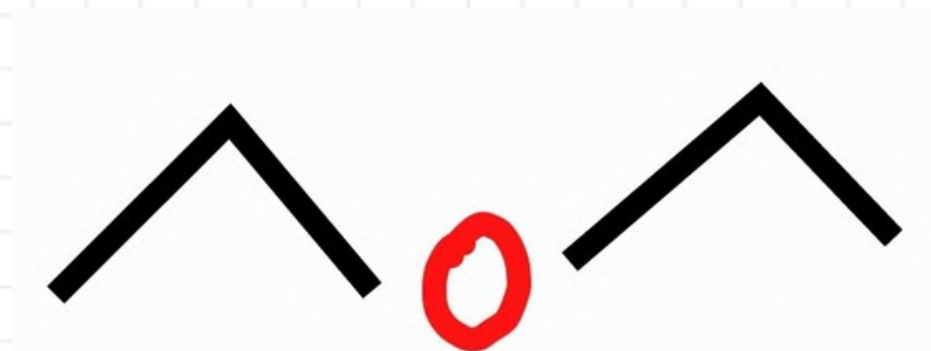
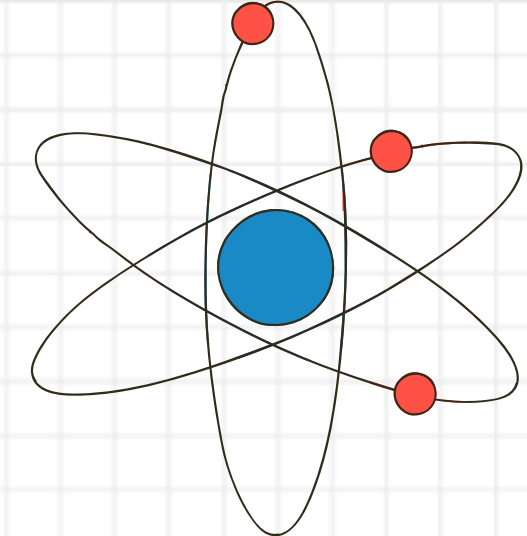


ÁCIDO CARBOXILÍCO

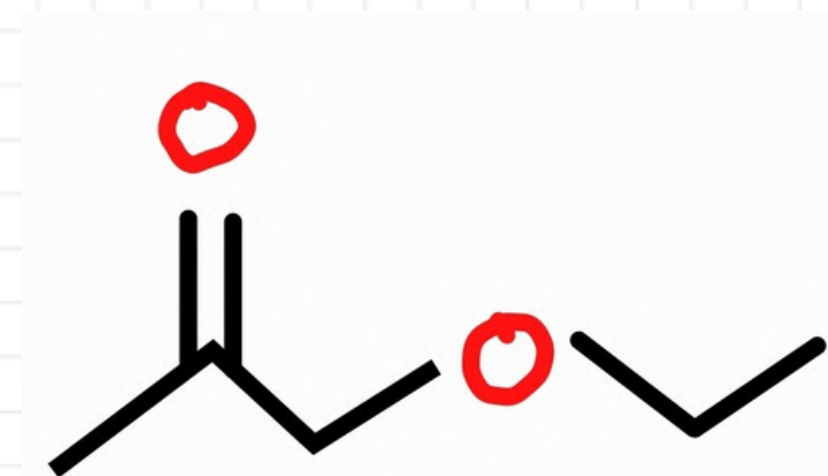


Revisão

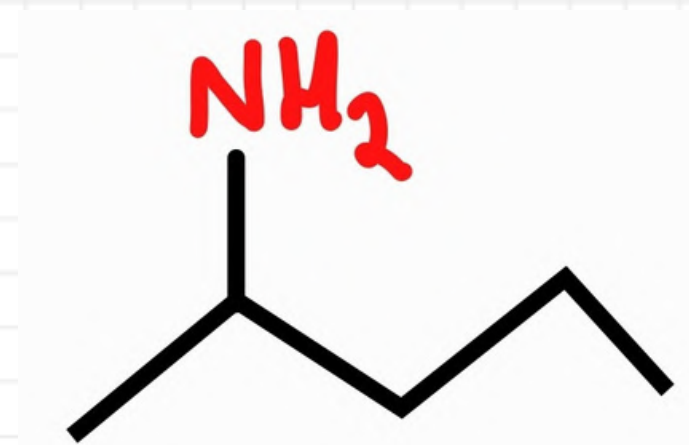
Grupos Funcionais



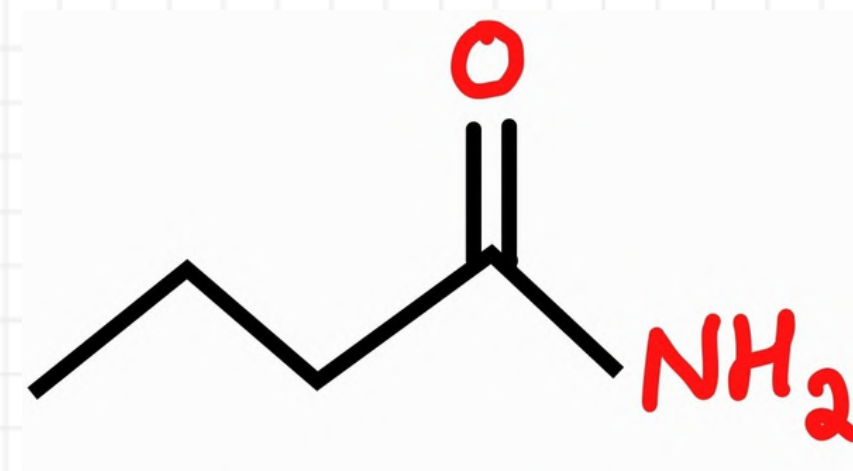
ÉTER



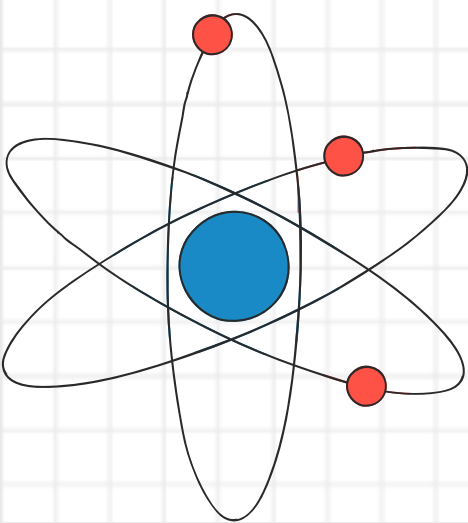
ÉSTER



AMINA

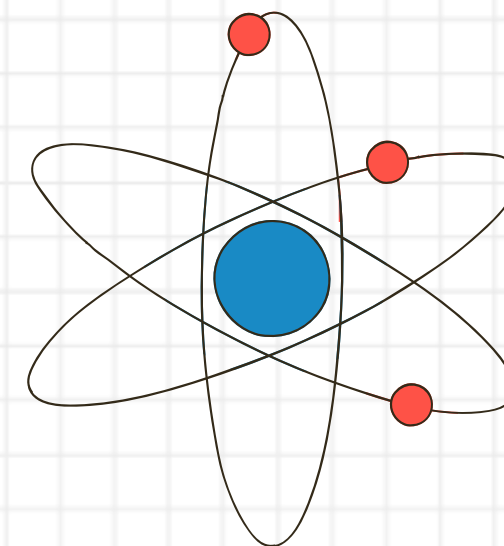
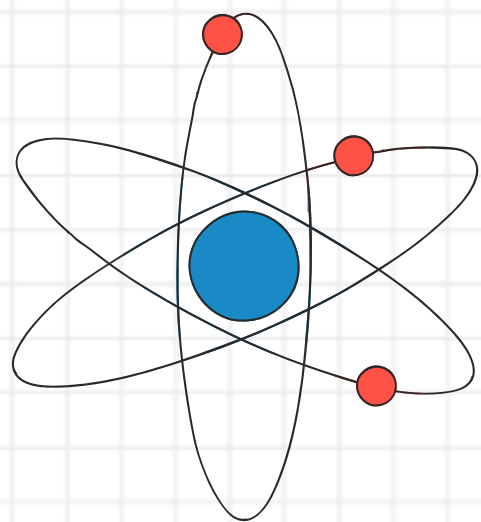


AMIDA

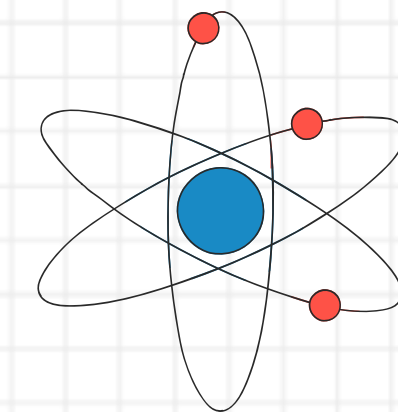


Como surgiu a Nomenclatura?

Estabelece regras para nomear e identificar os compostos orgânicos

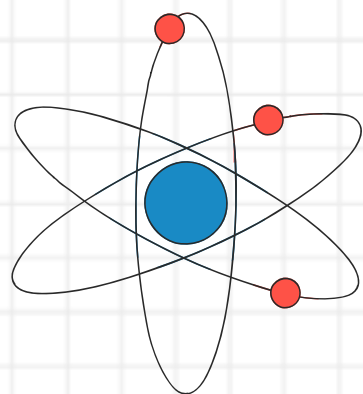
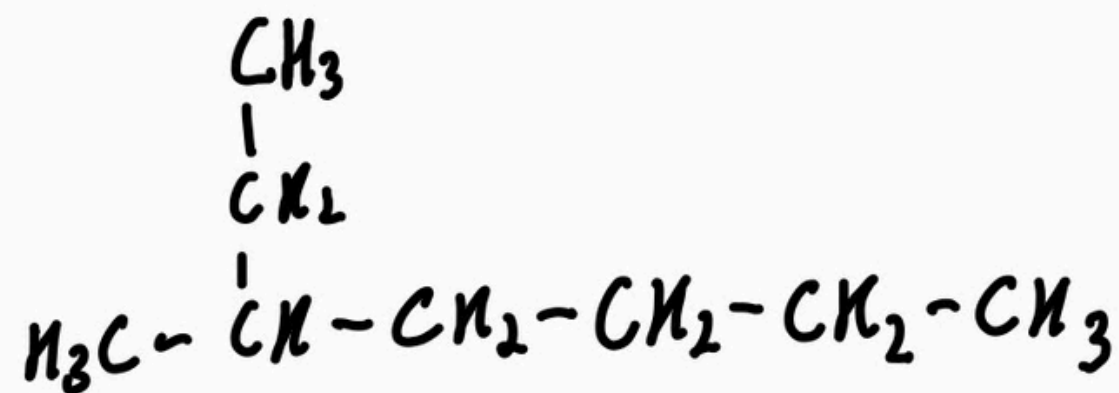
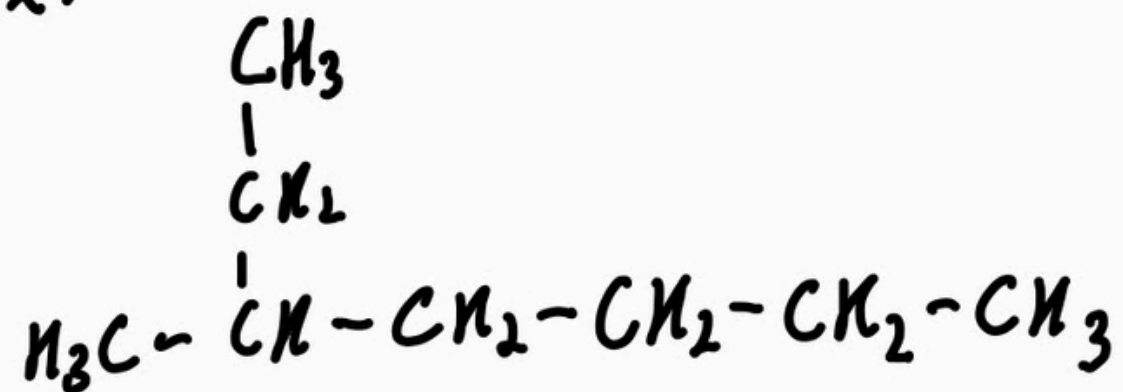


Regras para a Nomenclatura

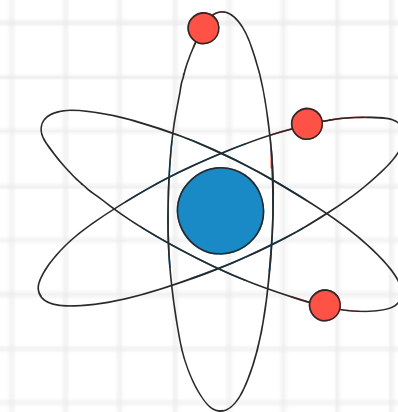


1º Identificação da Cadeia Principal

Ex:

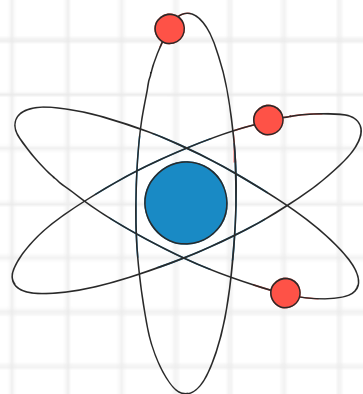
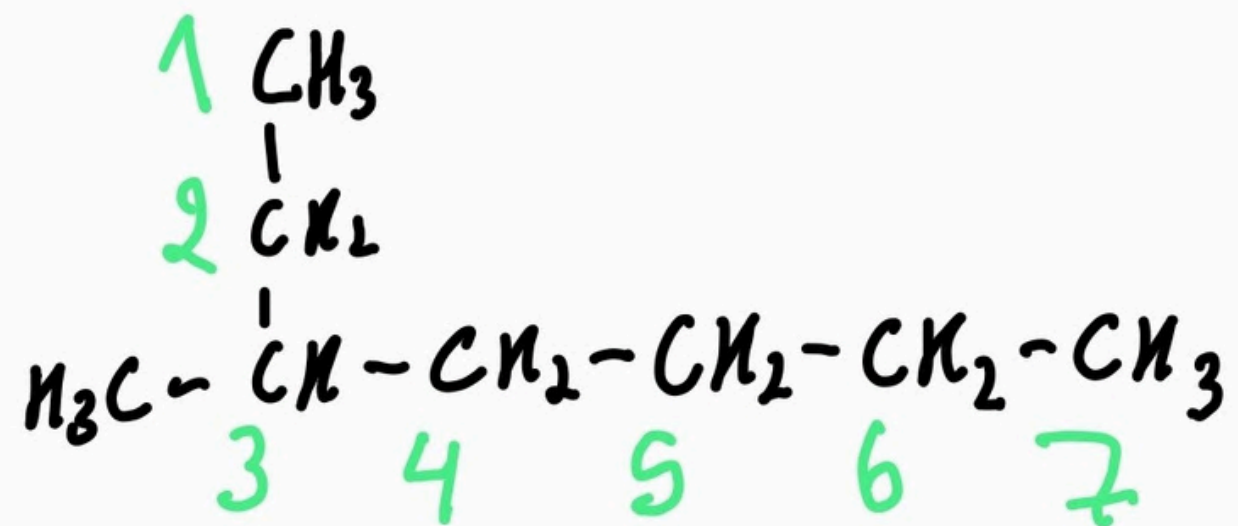
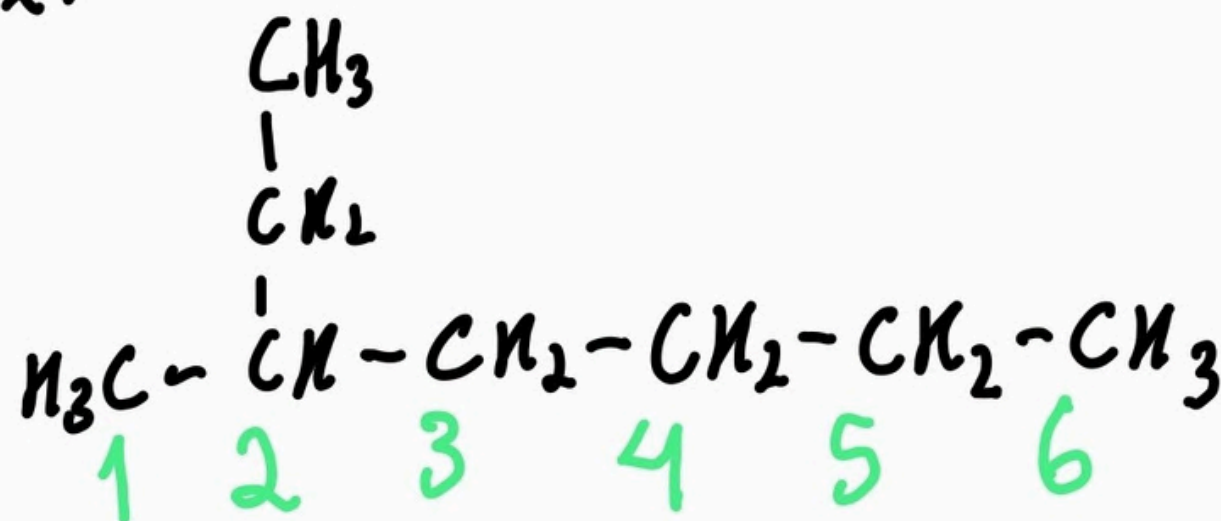


Regras para a Nomenclatura

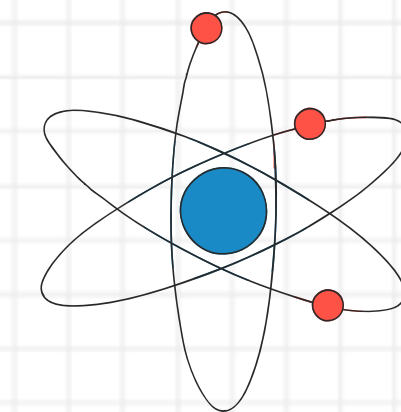


1º Identificação da Cadeia Principal

Ex:



Regras para a Nomenclatura

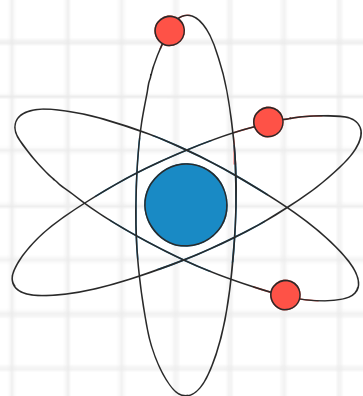
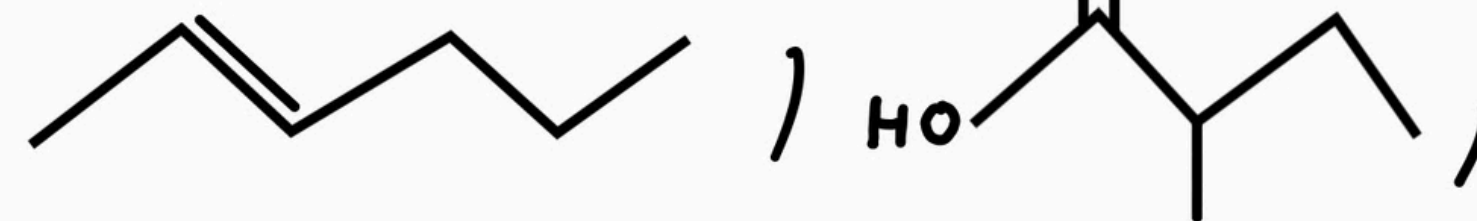


1º Identificação da Cadeia Principal

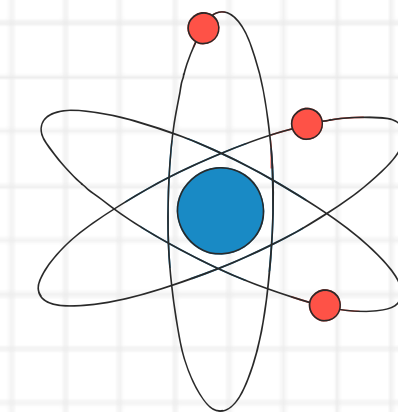
2º Numerar os Carbonos

Grupo Funcional > Insaturações > Ramificações

Ex:



Regras para a Nomenclatura

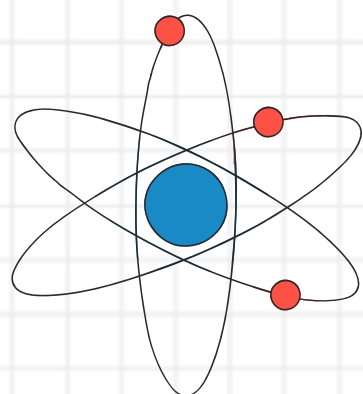
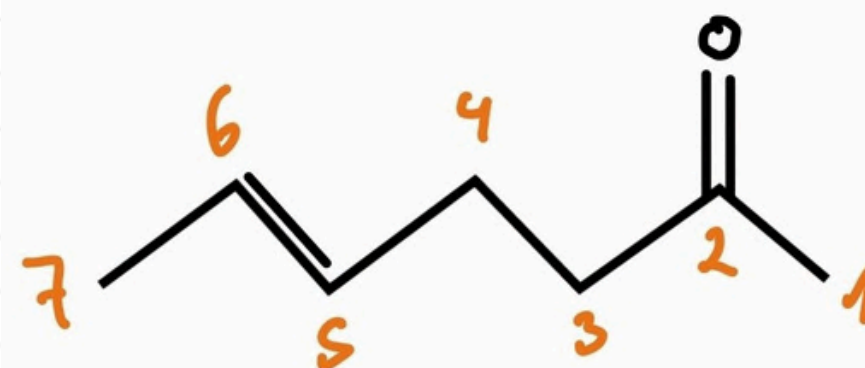
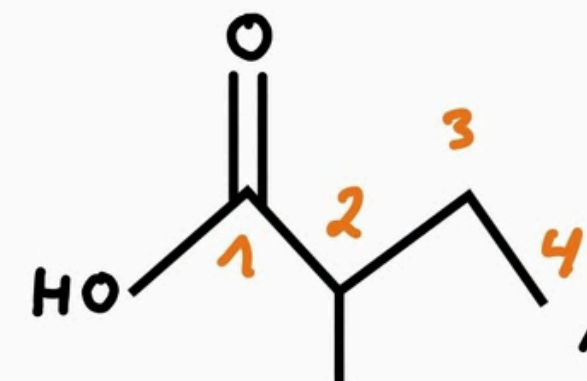
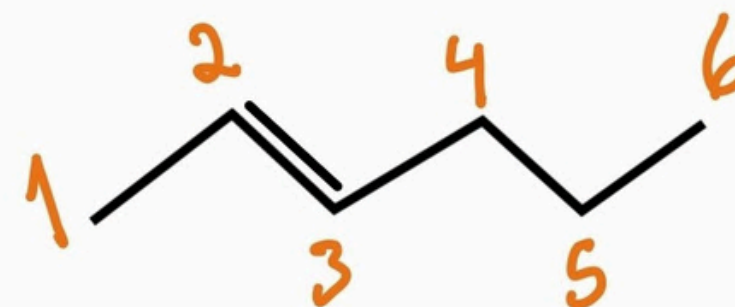


1º Identificação da Cadeia Principal

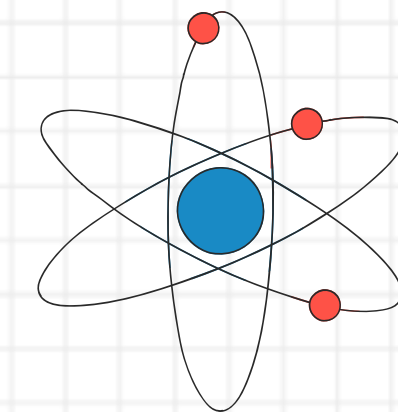
2º Numerar os Carbonos

Grupo Funcional > Insaturações > Ramificações

Ex:



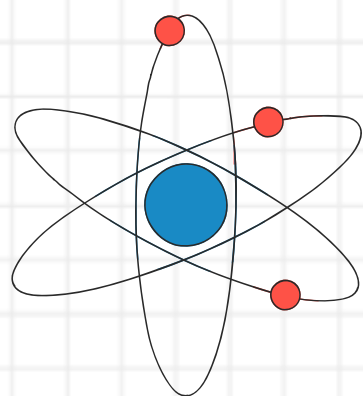
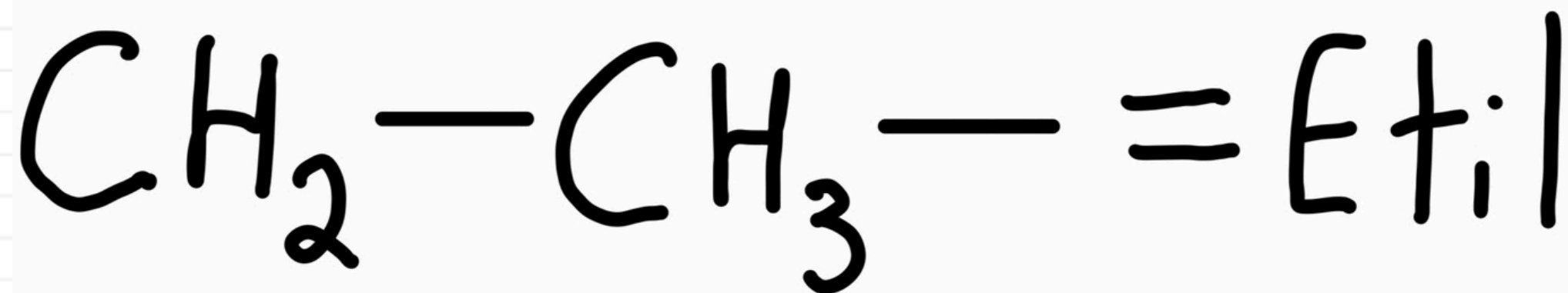
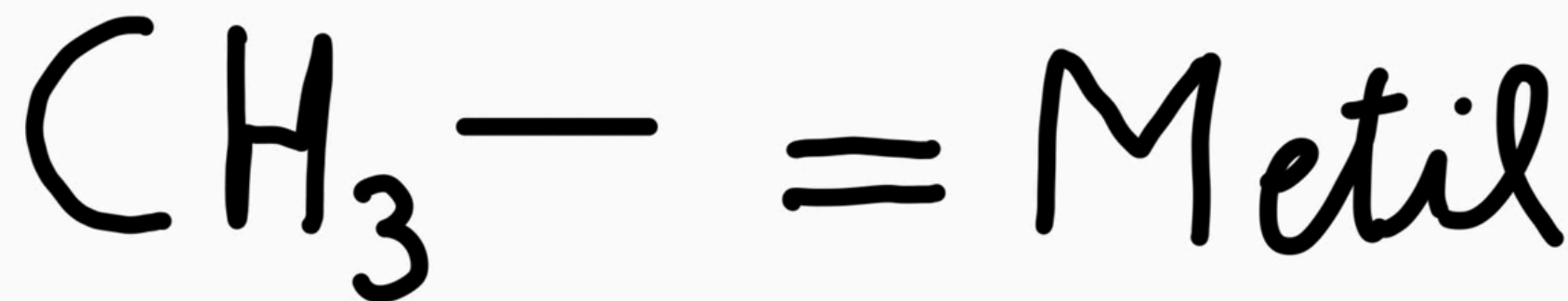
Regras para a Nomenclatura



1º Identificação da Cadeia Principal

2º Numerar os Carbonos

3º Identificar, numerar e nomear os substituintes (Radicais)

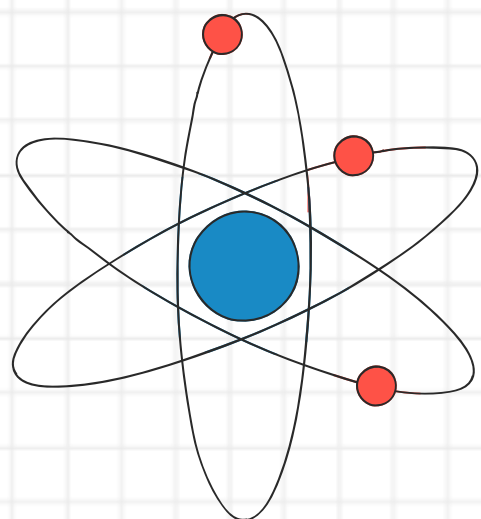
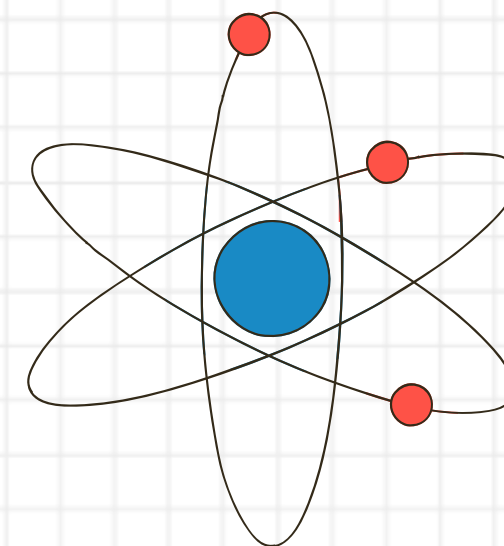


Estrutura da Nomenclatura

Ramificações + Cadeia Principal

↓
Ordem Alfabética

↓
Prefixo + Sufixo + Infixo



NOMENCLATURA

Ramificação

METIL
ETIL
(EM ORDEM
ALFABETICA)

Prefixo

1C- MET
2C- ET
3C- PROP
4C- BUT
5C- PENT
6C- HEX
7C- HEPT
8C- OCT
9C- NON
10C- DEC
.....

Sufixo

Lig. Simples- AN
Lig. Dupla- EN
Lig. Tripla- IN
Duas duplas- DIEN

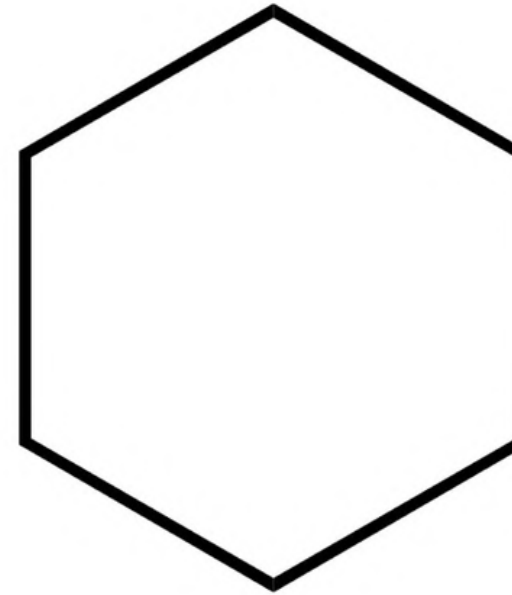
Infixo

HIDROCARBONETO- O
ÁLCOOL- OL
FENOL- FENOL
ALDEÍDO- AL
CETONA- ONA
ÁCIDO CARBOXÍLICO-
OICO
ÉTER- ÓXI + ANO
ÉSTER- OATO + ILA
AMINA- AMINA
AMIDA- AMIDA

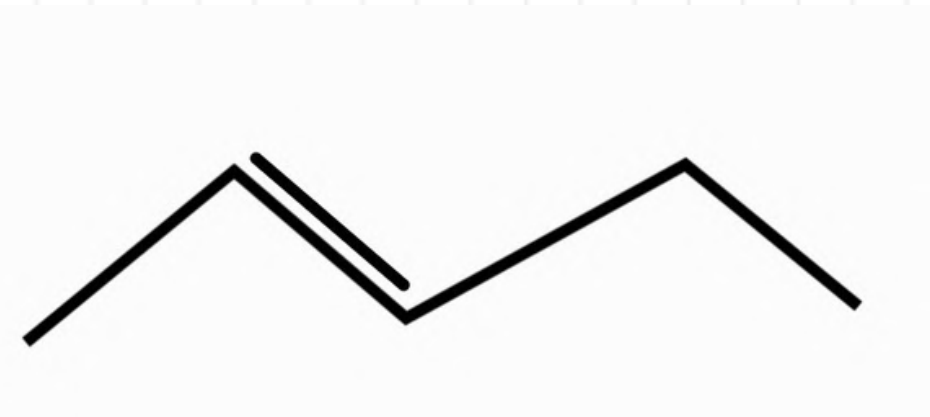
Hidrocarbonetos



BUTANO



Ciclo Hexano

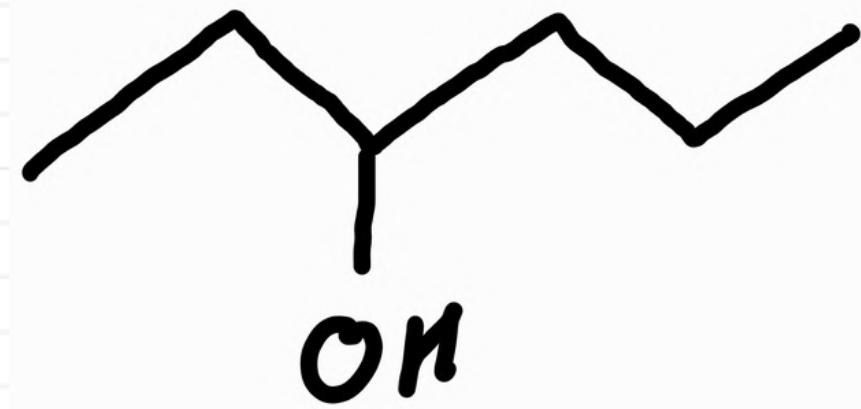


PENT-2-ENO

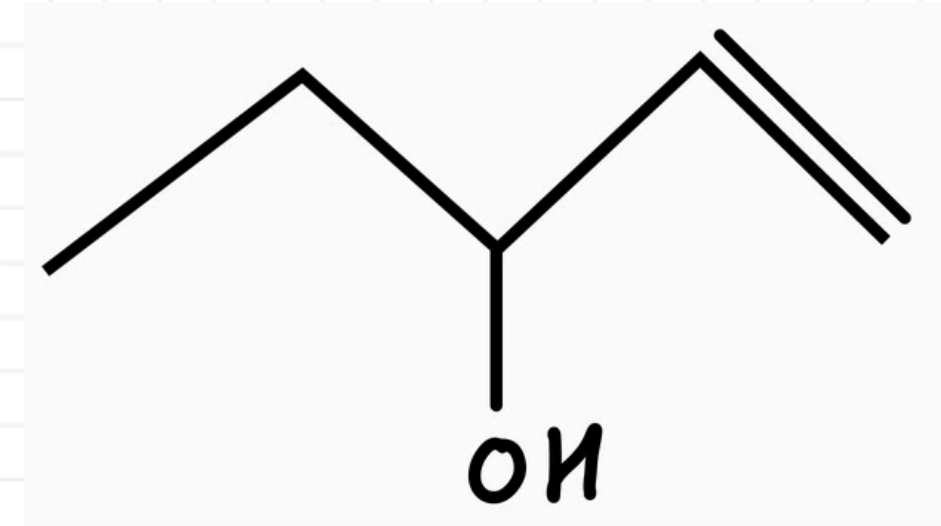


HEX-3-INO

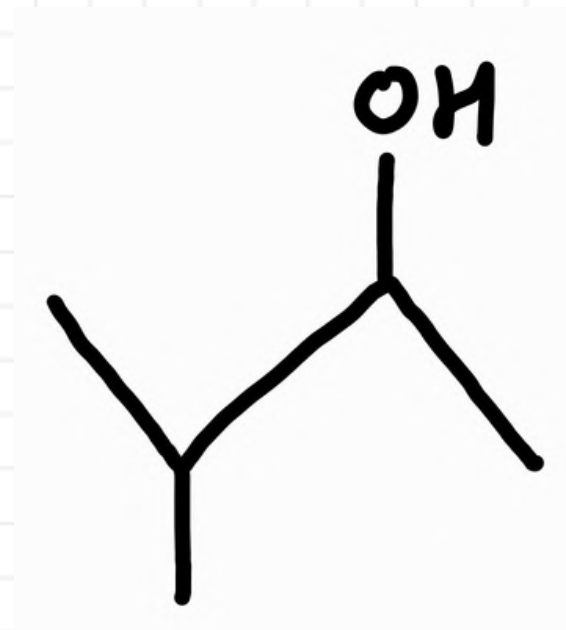
Álcool



HEXAN-3-OL

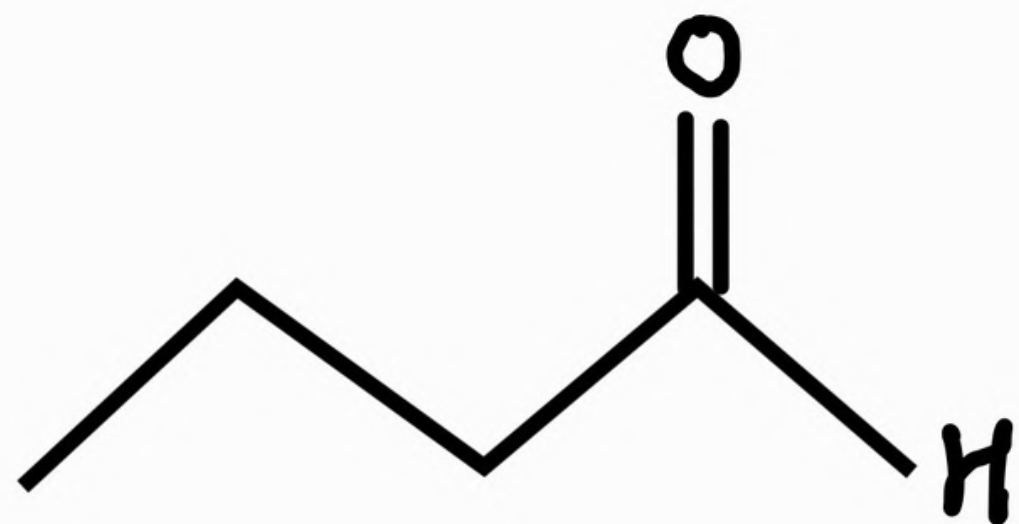


PENT-4-EN-3-OL

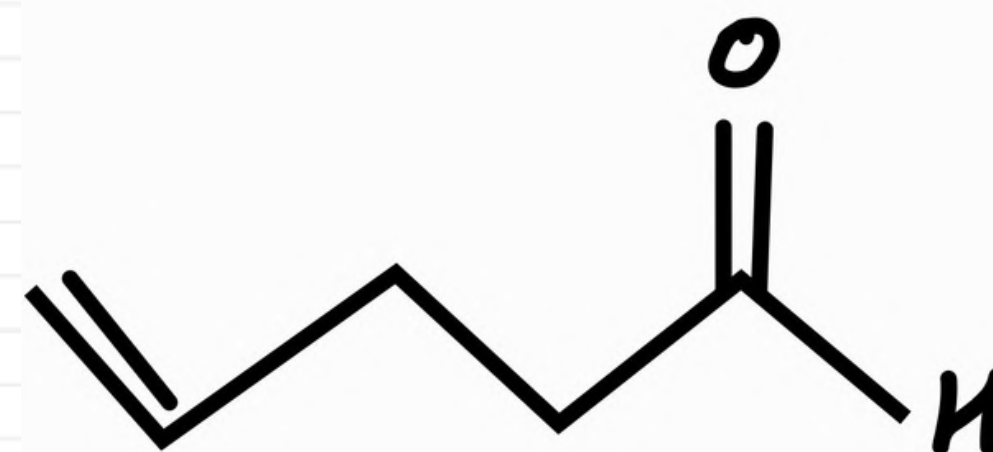


4-METIL-BUTAN-2-OL

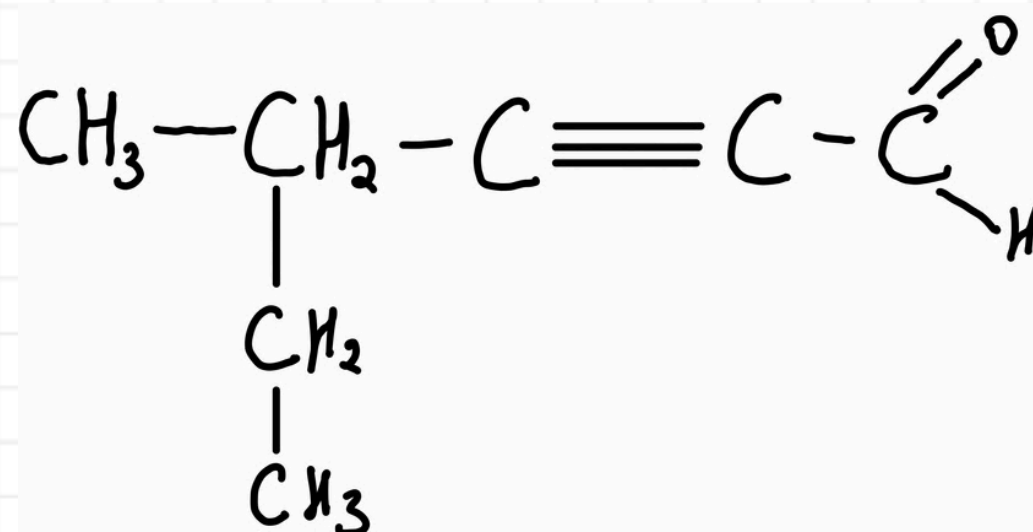
Aldeído



BUTANAL

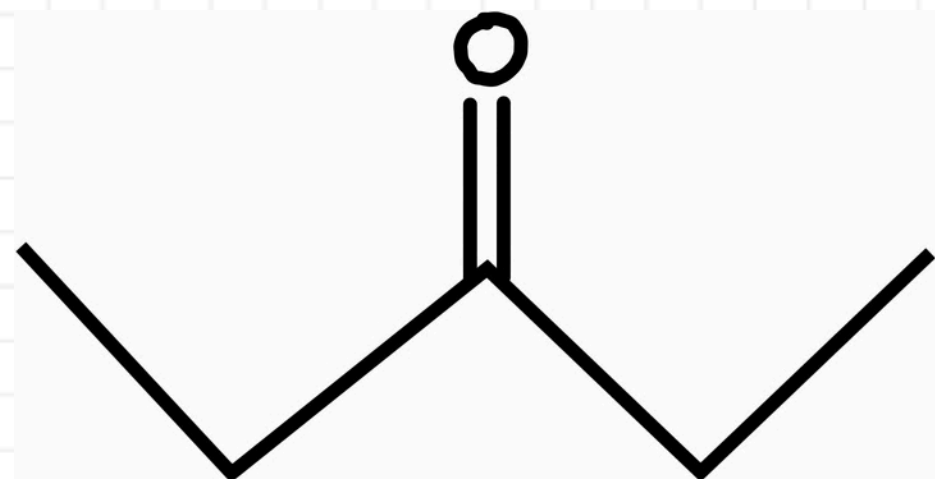


PENT-4-ENAL

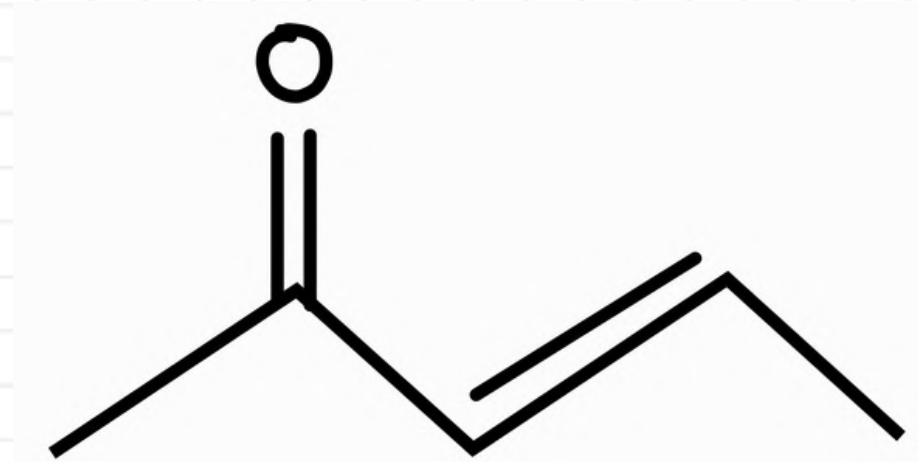


4-METIL-HEX-2-INAL

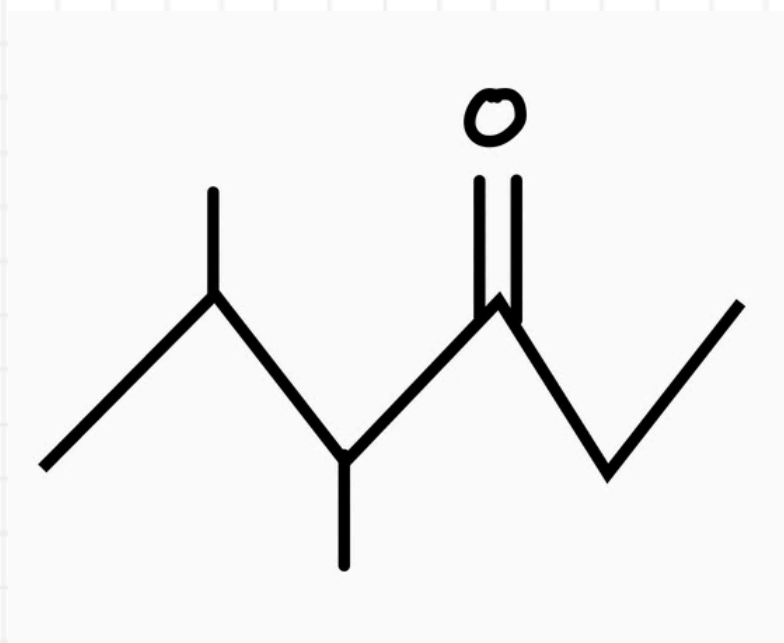
Cetona



PENTAN-3-ONA

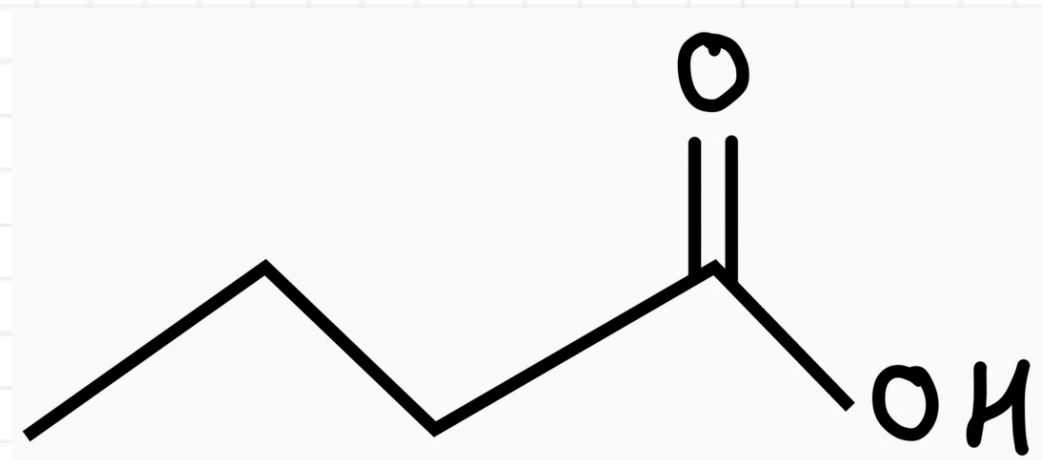


PENT-3-EN-2-ONA

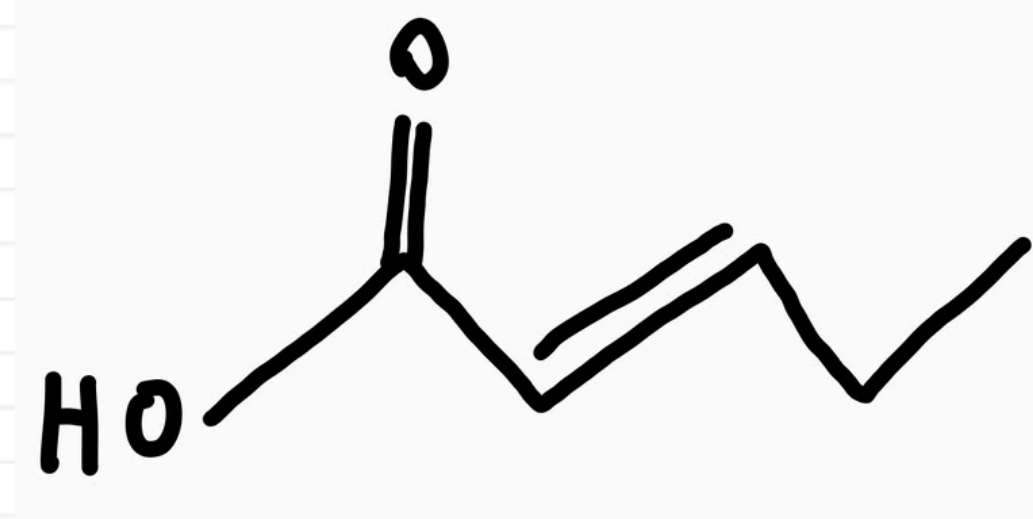


3,4-METIL-HEXAN-3-ONA

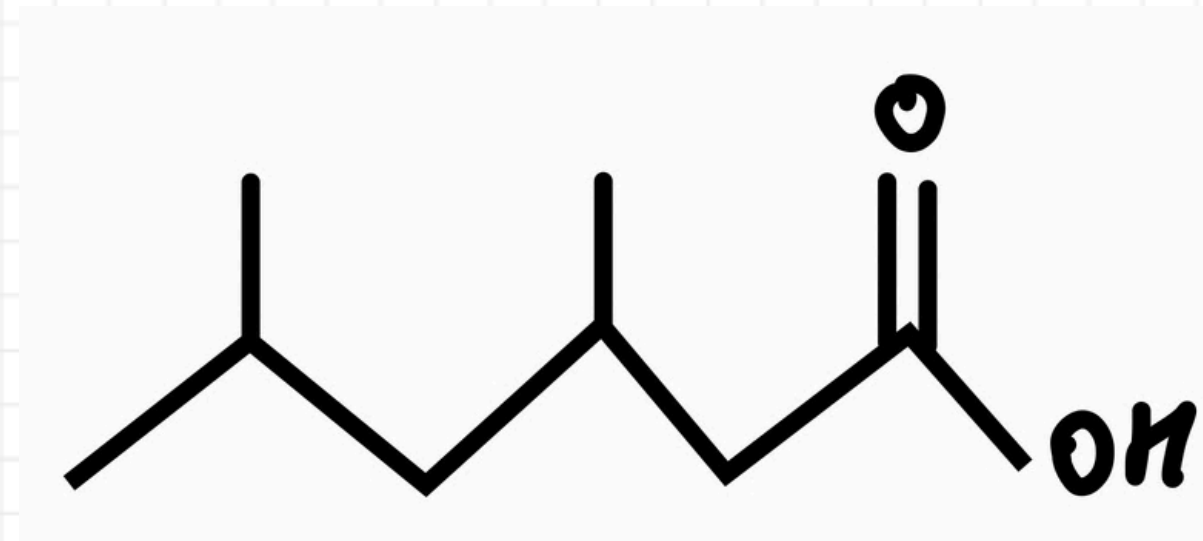
Ácido Carboxílico



ÁCIDO BUTANOICO



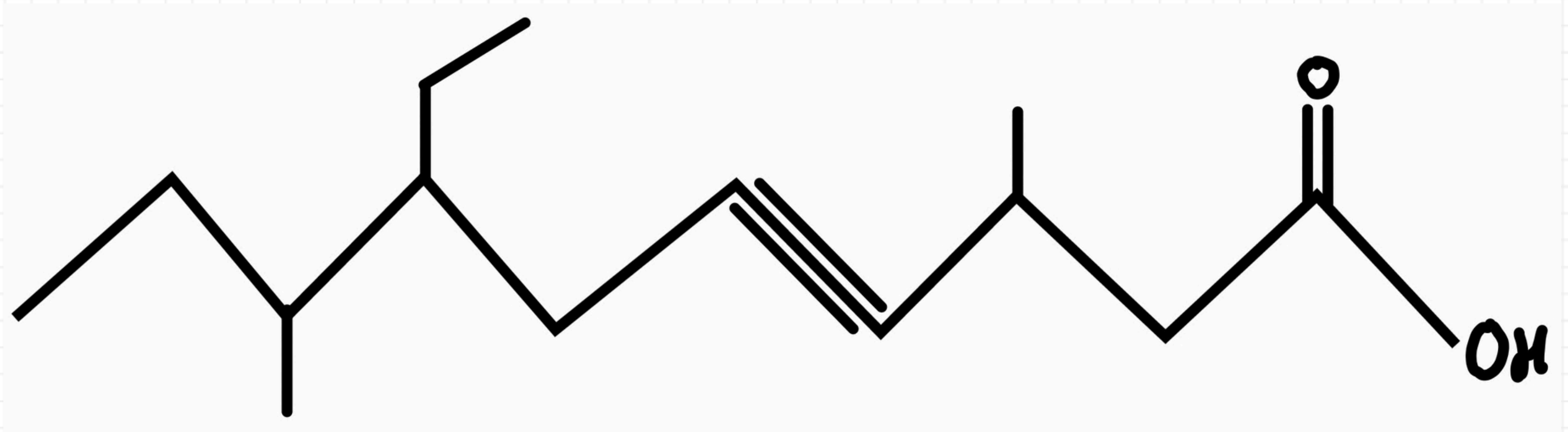
ÁCIDO PENT-3-ENOICO



ÁCIDO 3,5-DIMETIL-HEXANOICO

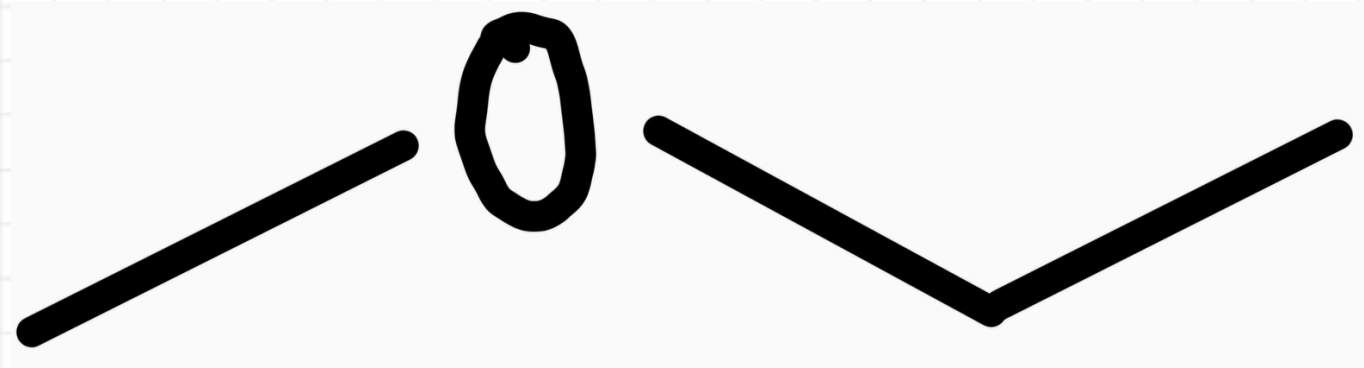
Ácido Carboxílico

DESAFIO!!!!

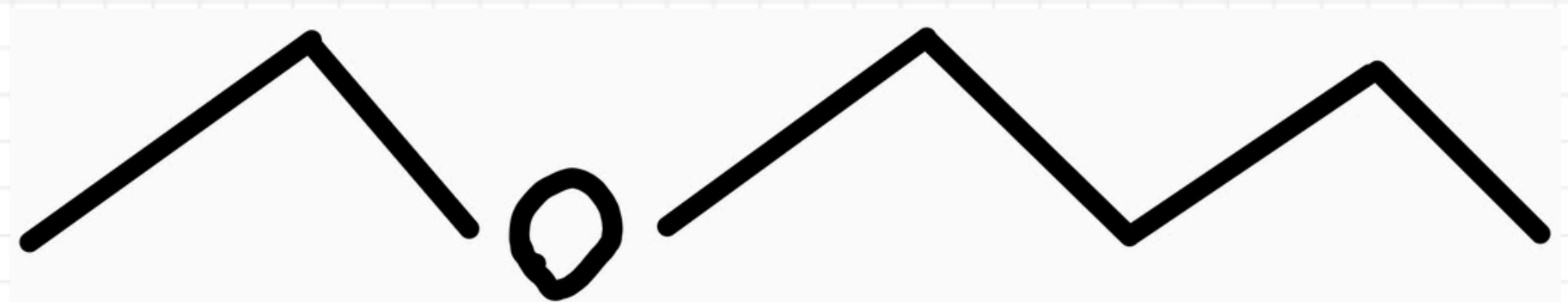


ÁCIDO 7-ETIL-3,8-DIMETIL-DEC-4-INOICO

Éter

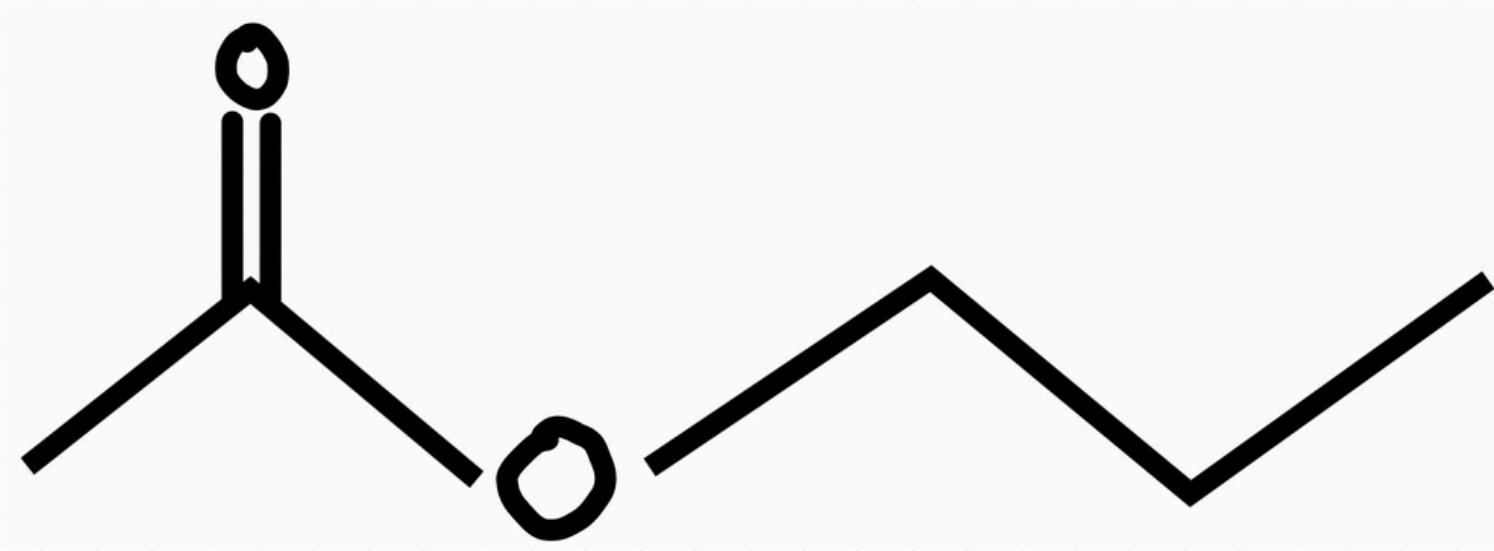


METOXIETANO

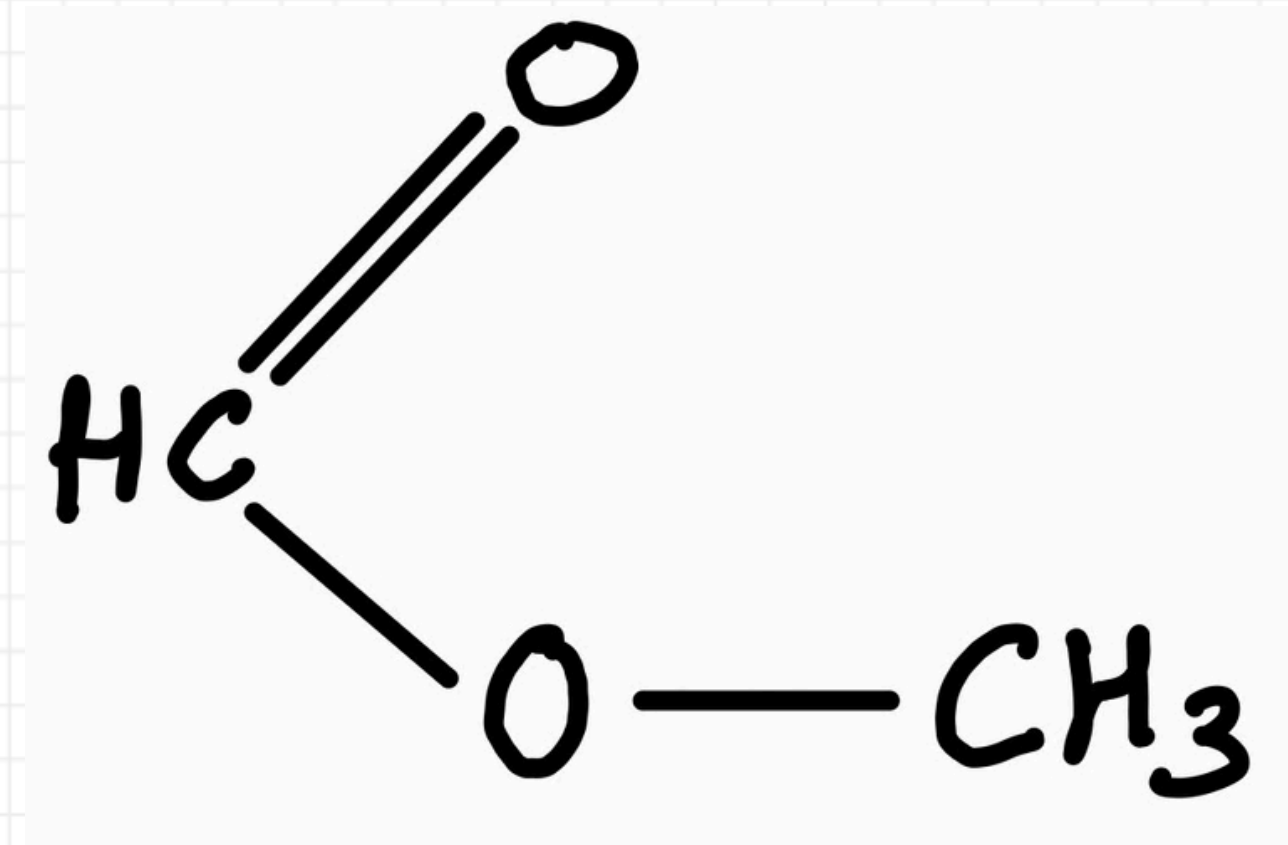


ETOXIBUTANO

Éster

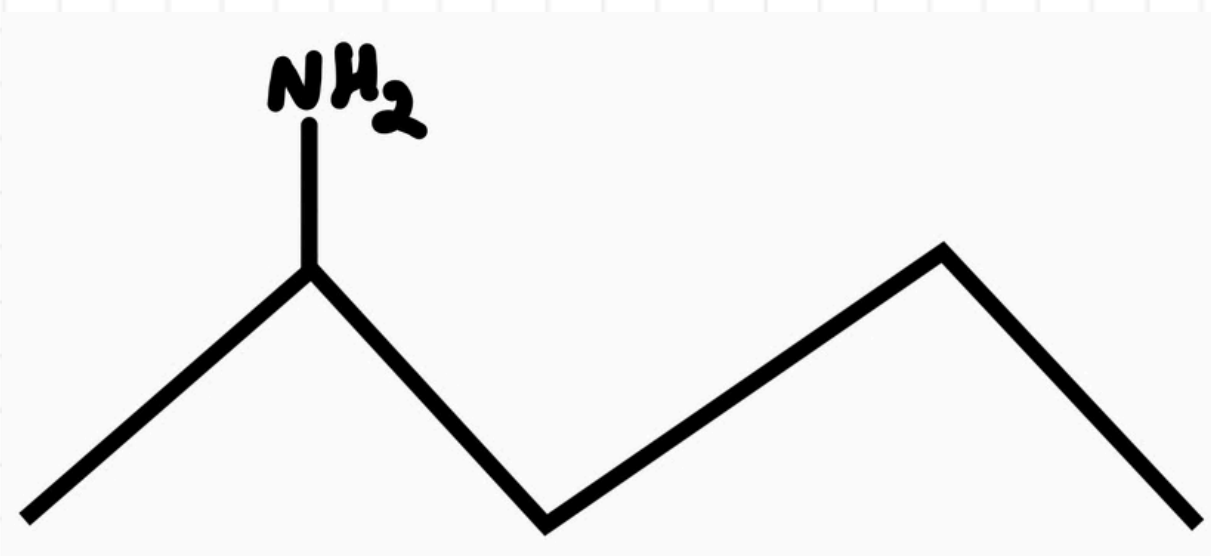


PROPANOATO DE ETILA

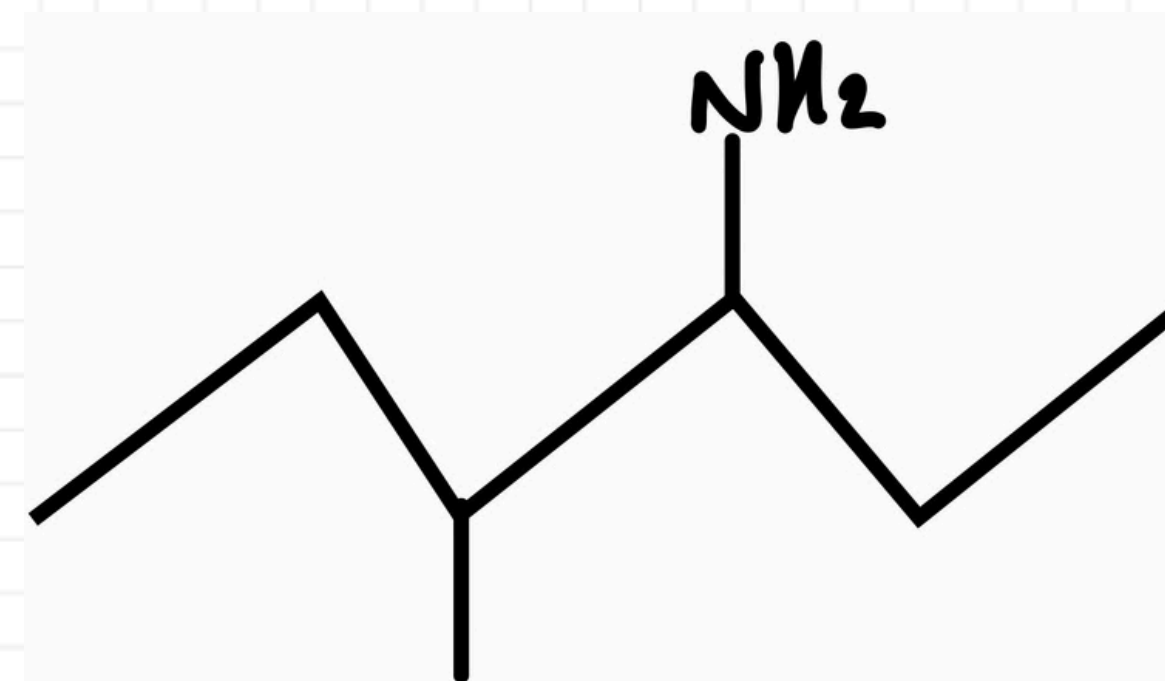


METANOATO DE METILA

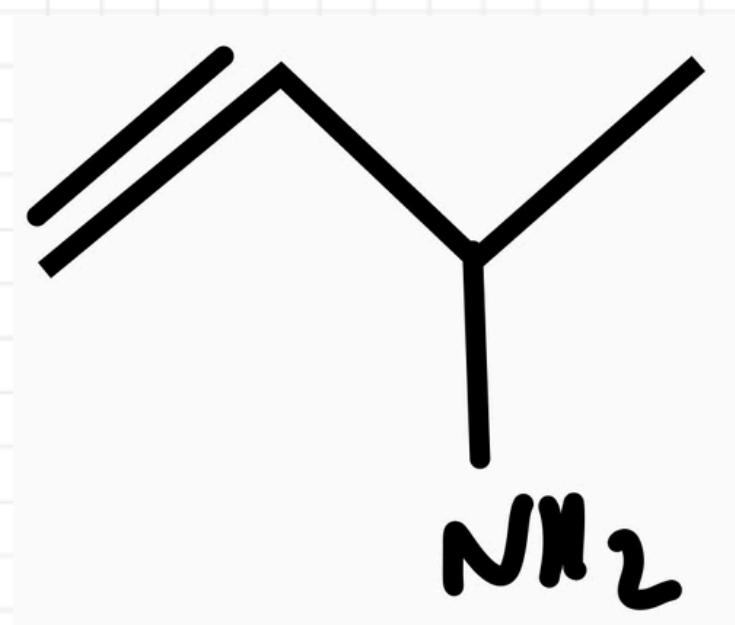
Amina



PENTAN-2-AMINA

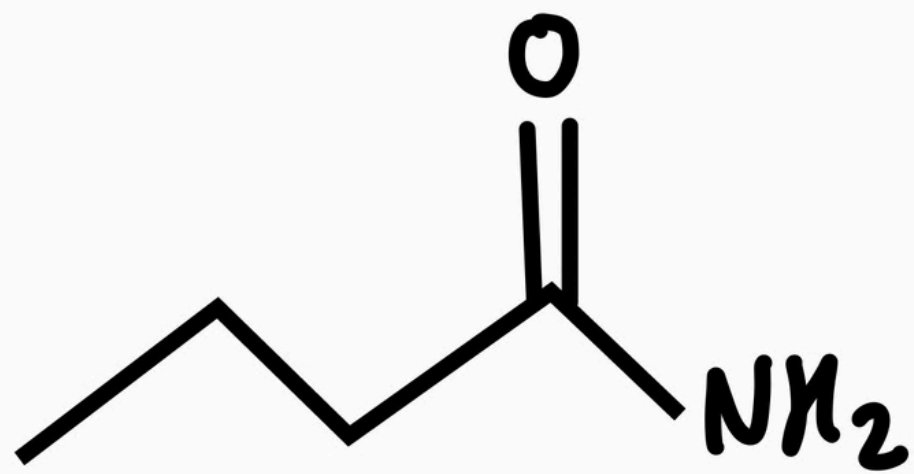


4-METIL-HEXAN-3-AMINA

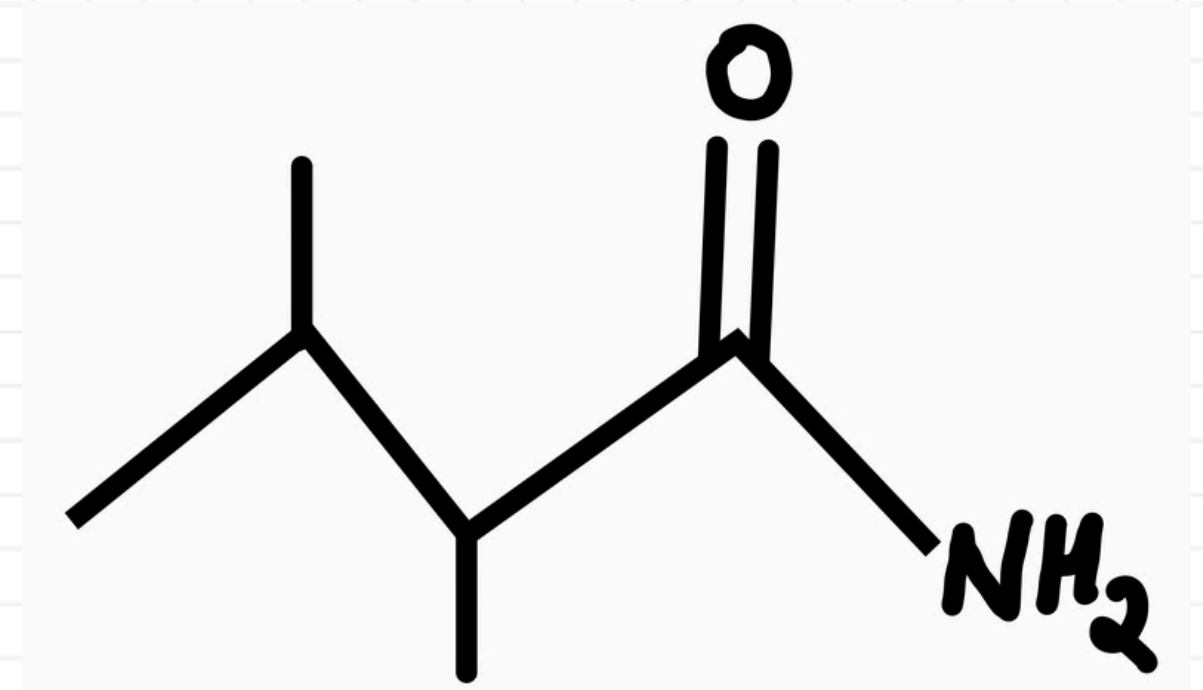


BUT-3-EN-2-AMINA

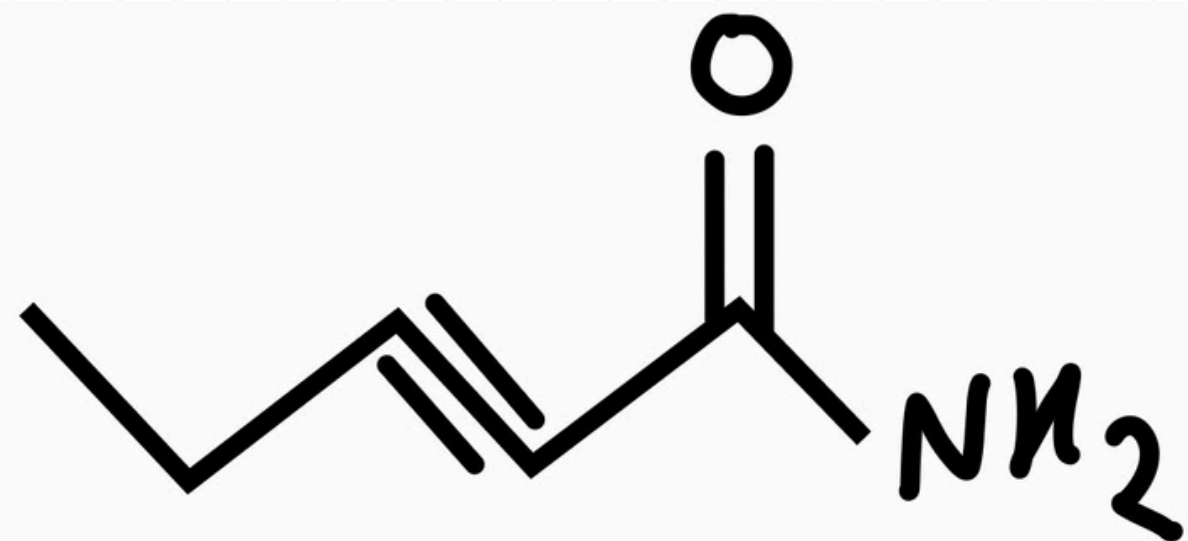
Amida



BUTANAMIDA



2,3-DIMETIL-BUTANAMIDA



PENT-2-INAMIDA

Exercícios

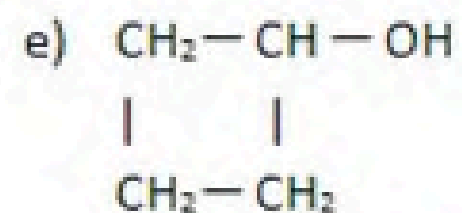
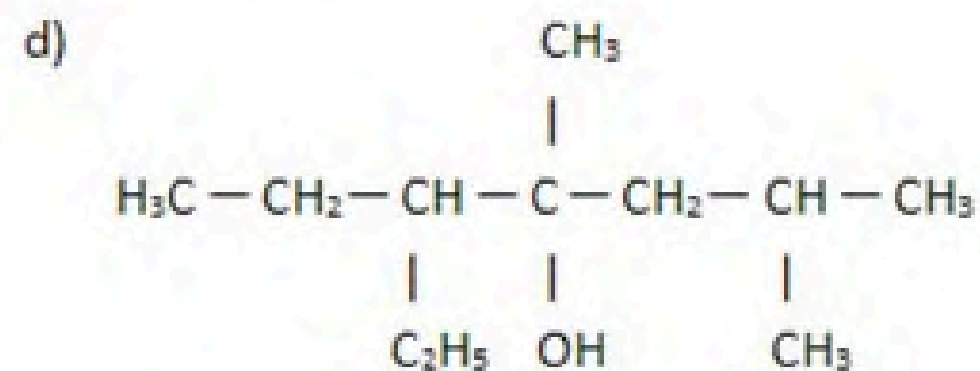
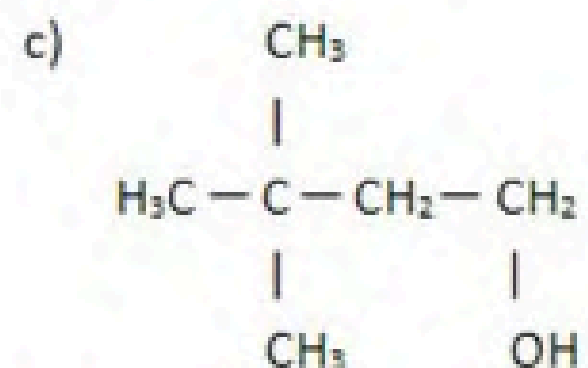
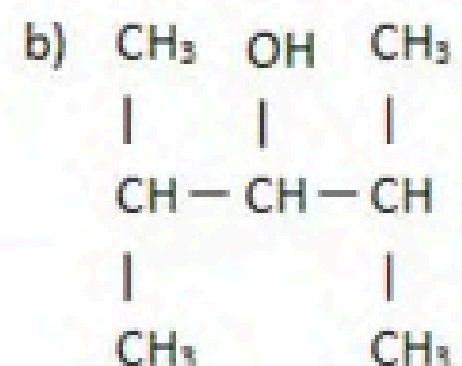
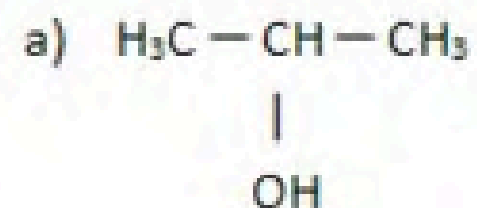
Na Química orgânica, os compostos são reconhecidos pelas cadeias formadas por carbono e hidrogênio. Entretanto, outros elementos podem fazer parte da estrutura química desses compostos, como o oxigênio.

Selecione a alternativa em que os dois compostos orgânicos apresentam funções orgânicas oxigenadas.

- a) clorofórmio e metanoato de etila
- b) propanol e ácido propanoico
- c) eteno e etanodiol
- d) etanamida e benzeno

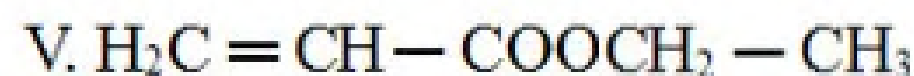
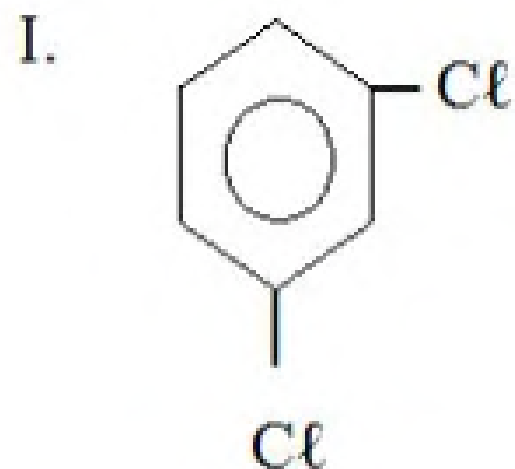
Exercícios

Dê os nomes, segundo as regras estabelecidas pela IUPAC (União Internacional de Química Pura e Aplicada), dos seguintes compostos:



Exercícios

(ENCE-UERJ-Cefet) Observe as fórmulas estruturais a seguir:



Segundo a ordem em que aparecem, sua nomenclatura oficial é:

- 1,3-diclorobenzeno, ácido etanoico, dietilamida, 1-buteno e propanoato de etila.
- ácido etanoico, 1,3-diclorobenzeno, dietilamina, 3-buteno e propenoato de etila.
- 2,4-diclorobenzeno, ácido etanoico, dietilamida, 2-buteno e etanoato de propenila.
- 1,3-diclorobenzeno, ácido etanoico, dietilamina, 1-buteno e propenoato de etila.
- 2,4-diclorobenzeno, ácido etanoico, dietilamina, 2-buteno e propanoato de etila.



Muito

Obrigado