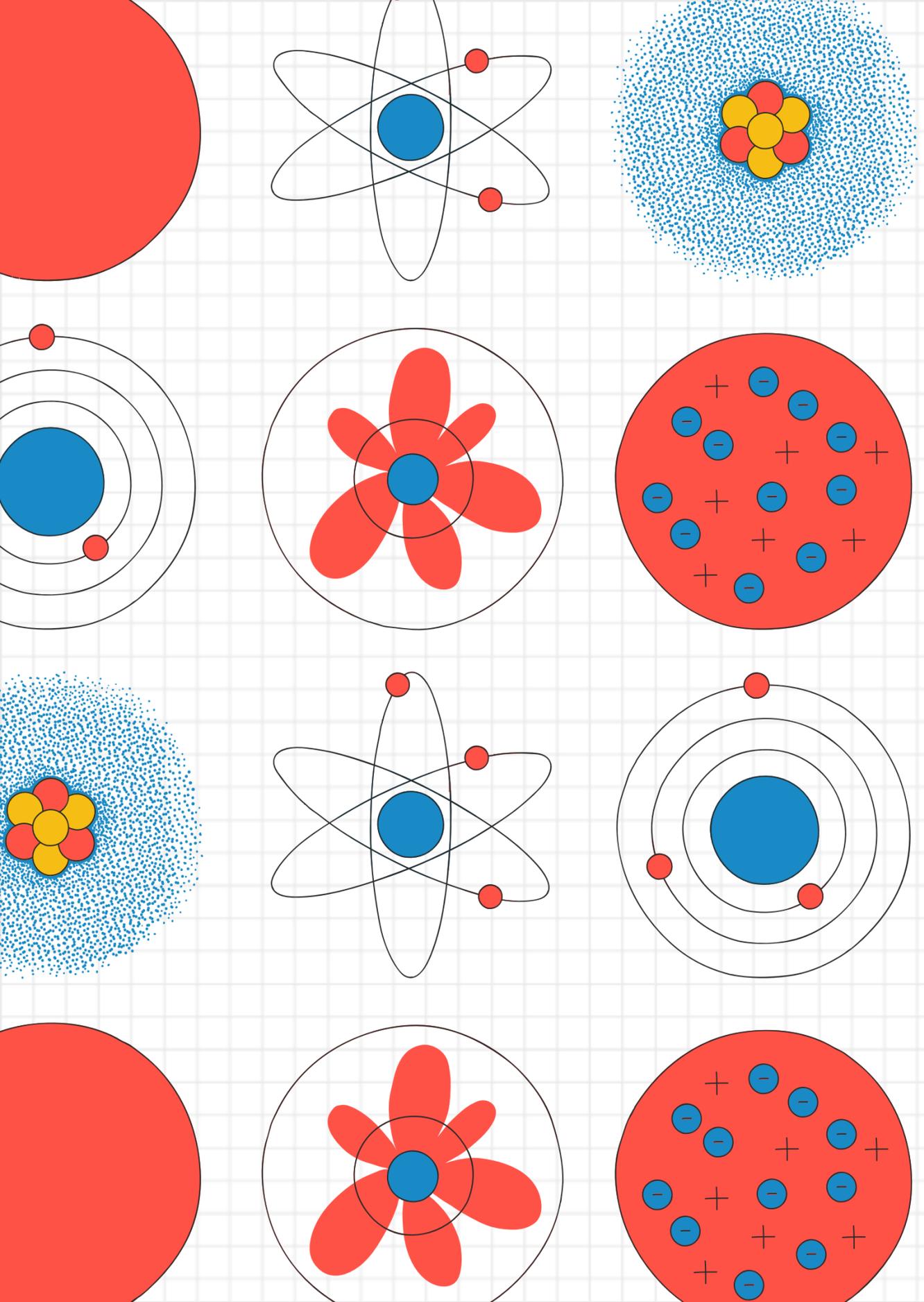




# Nomenclatura de Compostos Orgânicos

Prof. Pedro



# Na aula de Hoje

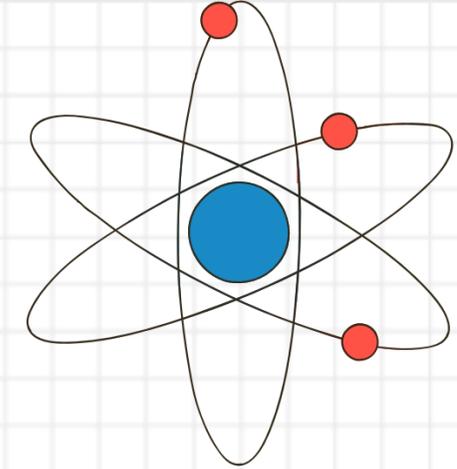
**Revisão**

**Como surgiu a nomenclatura**

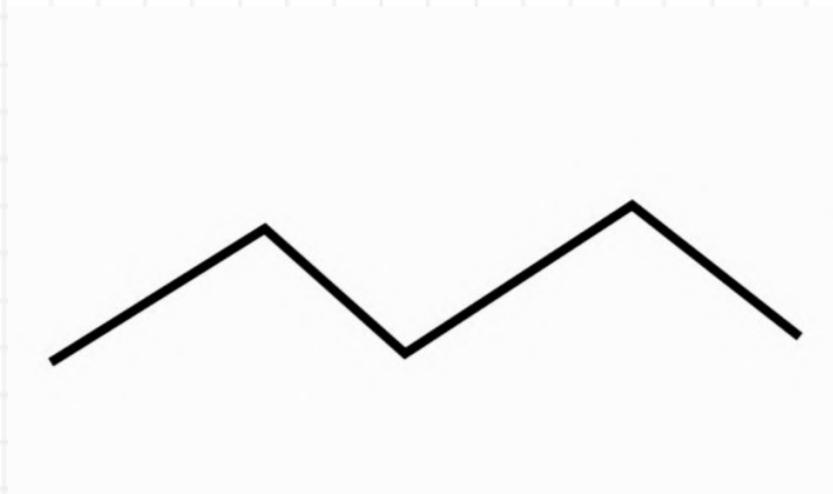
**Regras para a nomenclatura**

**Exercícios**

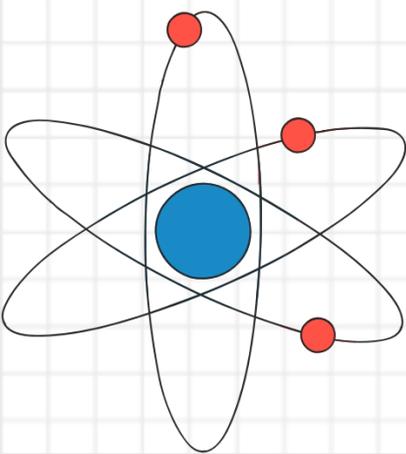
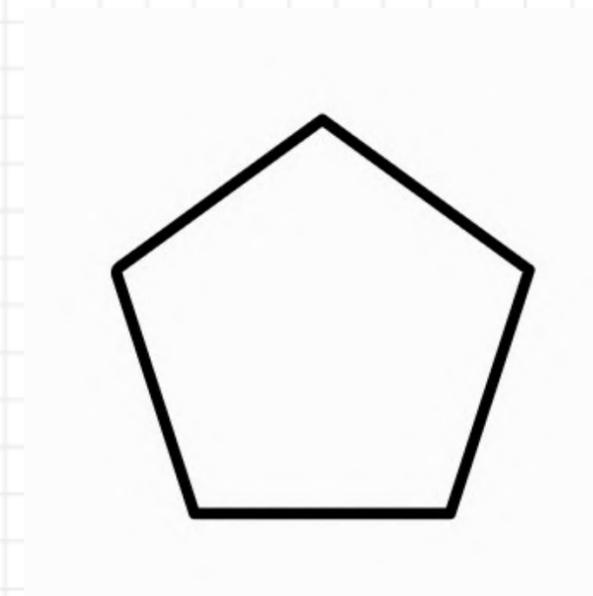
# Revisão



Cadeia Aberta

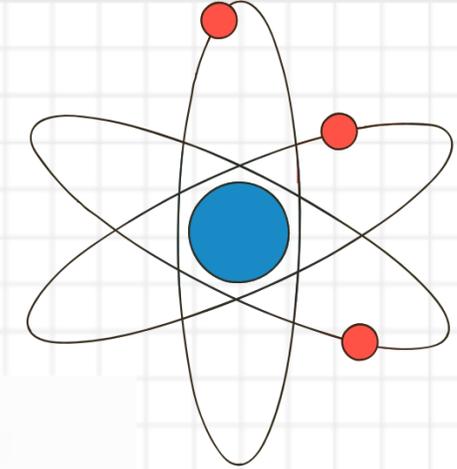


Cadeia Fechada



# Revisão

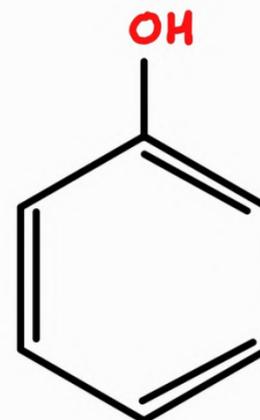
## Grupos Funcionais



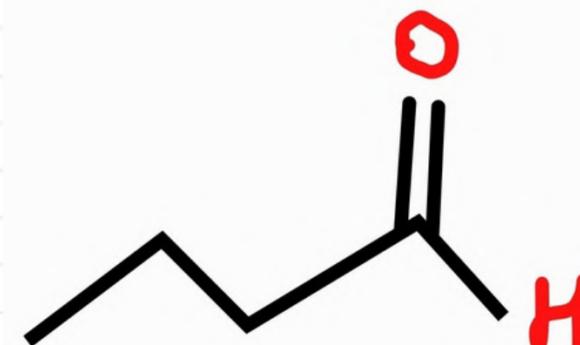
**HIDROCARBONETOS**



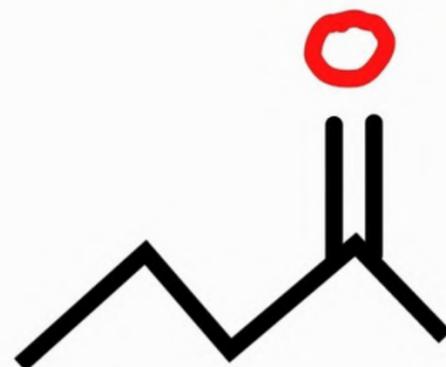
**ÁLCOOL**



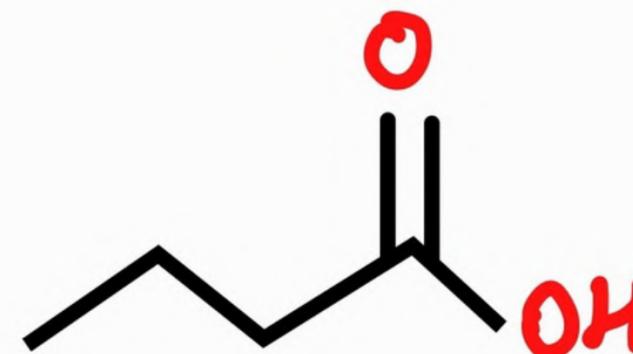
**FENOL**



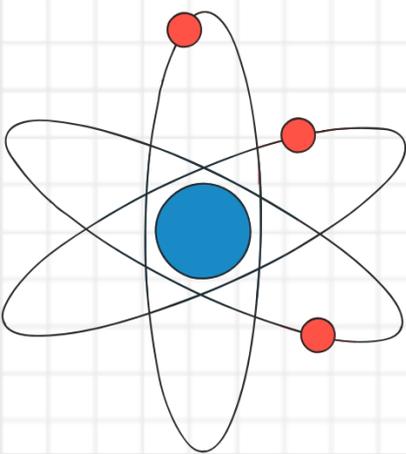
**ALDEÍDO**



**CETONA**

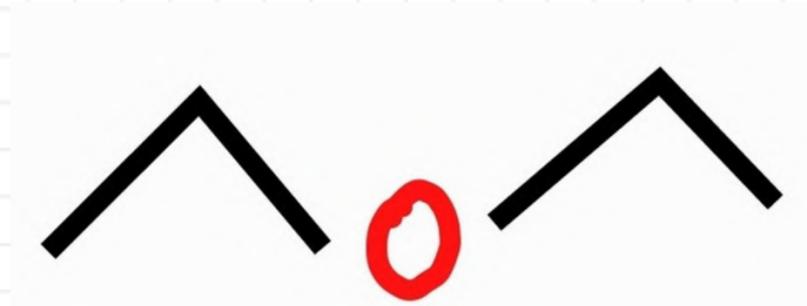
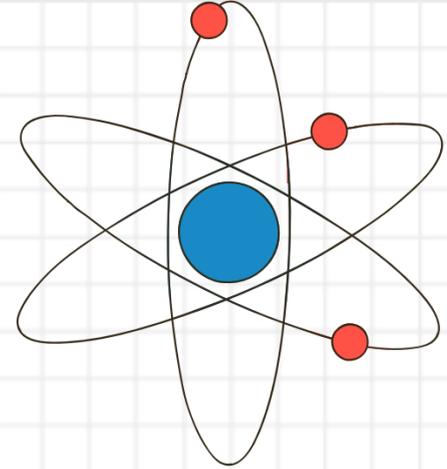


**ÁCIDO CARBOXILÍCO**

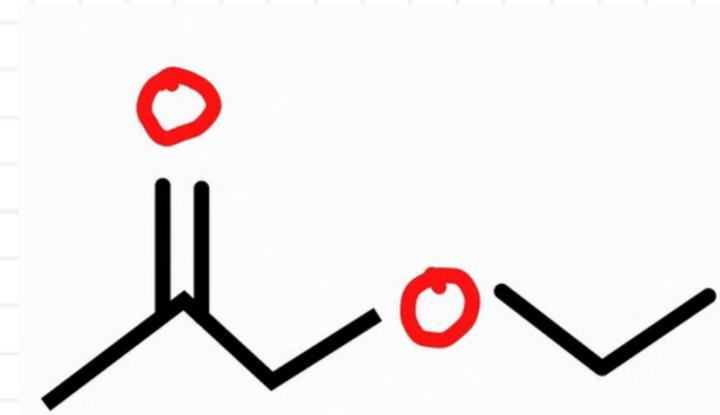


# Revisão

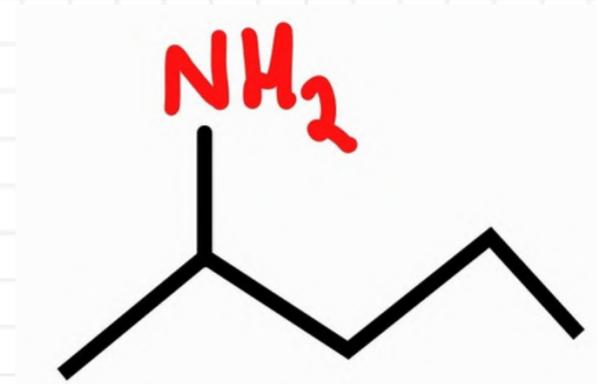
## Grupos Funcionais



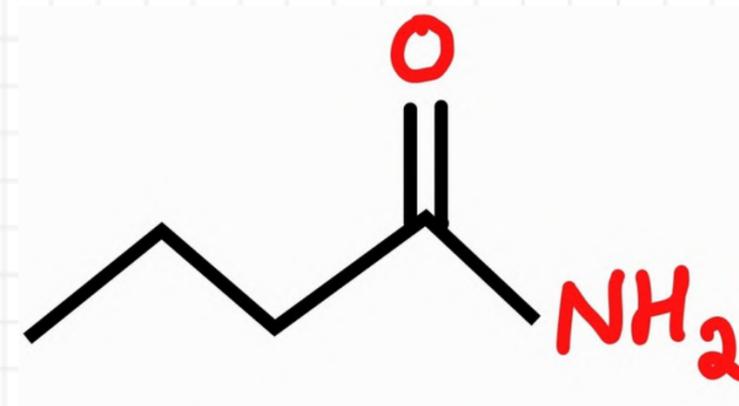
**ÉTER**



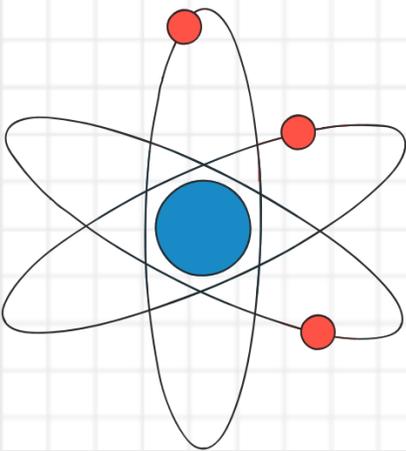
**ÉSTER**



**AMINA**

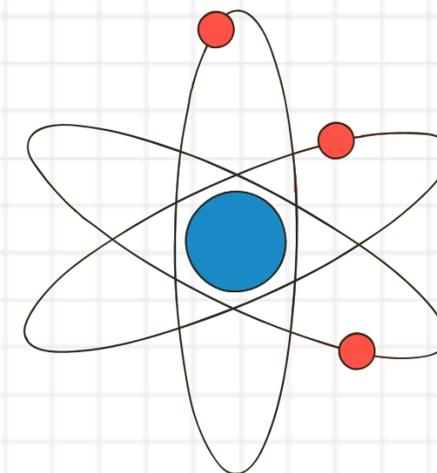
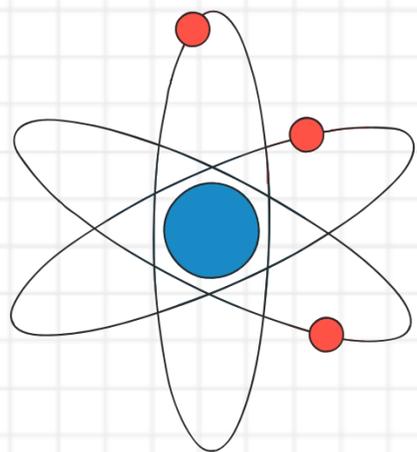


**AMIDA**

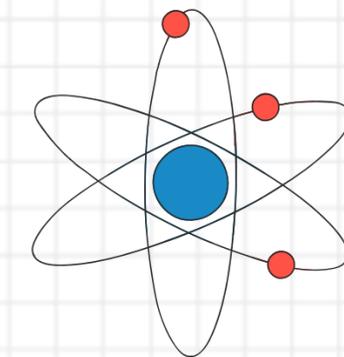


# Como surgiu a Nomenclatura?

Estabelece regras para nomear e identificar os compostos orgânicos

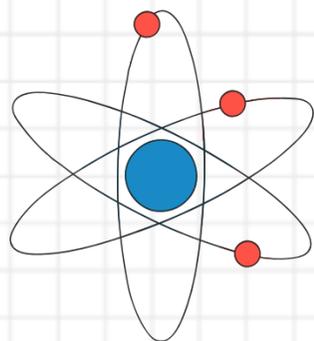
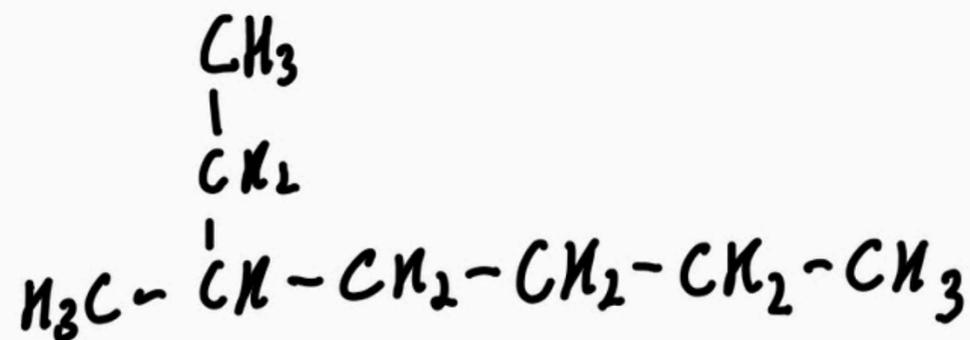
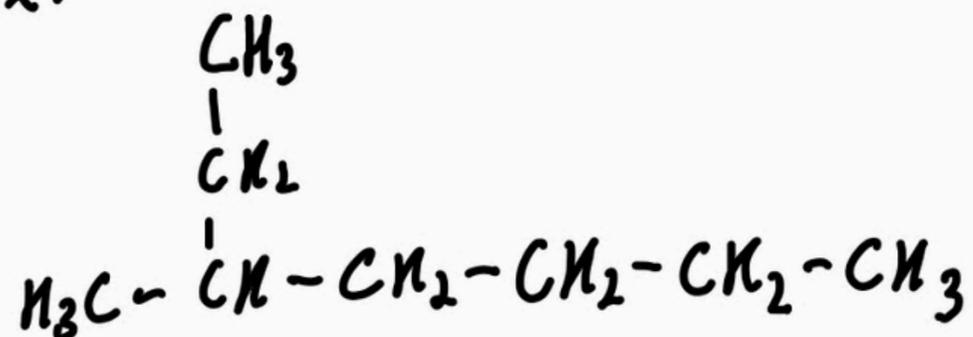


# Regras para a Nomenclatura

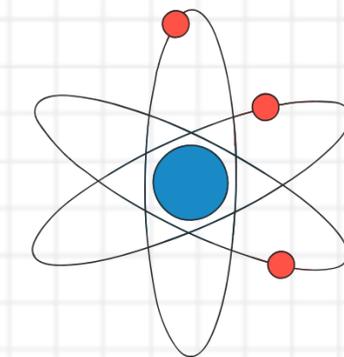


## 1º Identificação da Cadeia Principal

Ex:

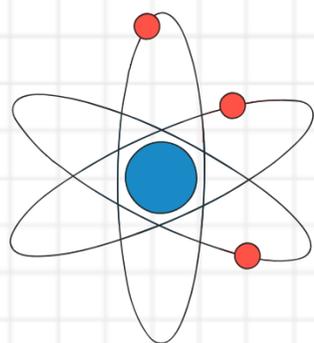
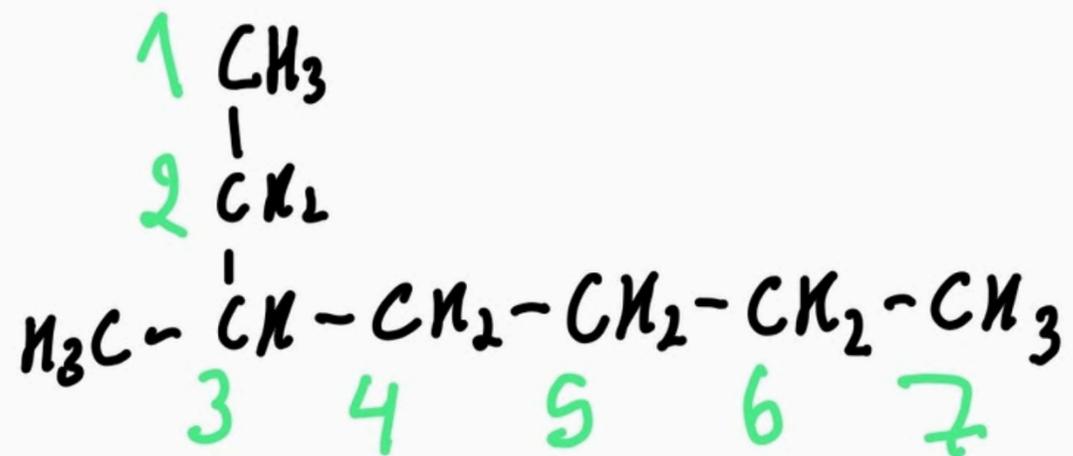
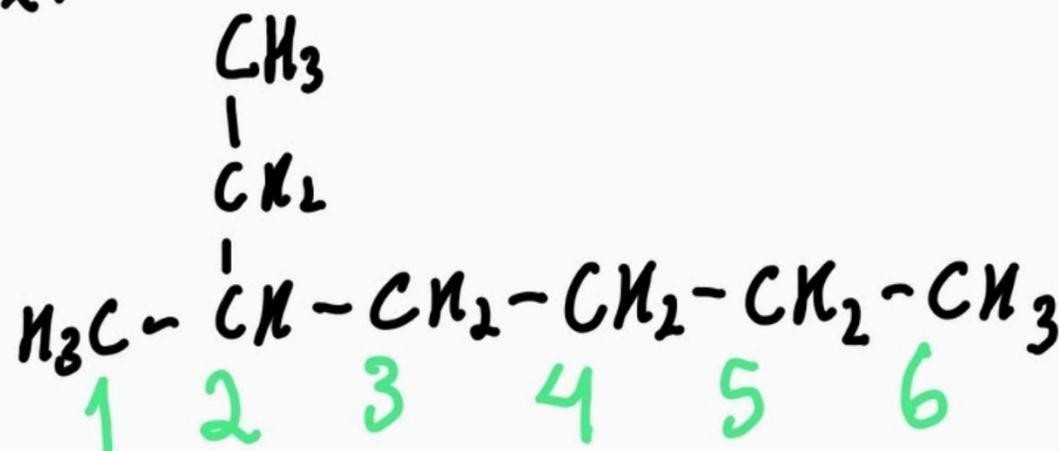


# Regras para a Nomenclatura

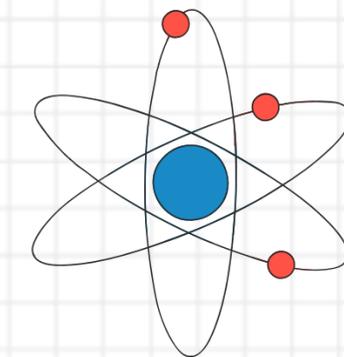


## 1º Identificação da Cadeia Principal

Ex:



# Regras para a Nomenclatura

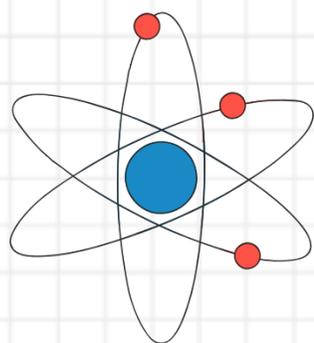
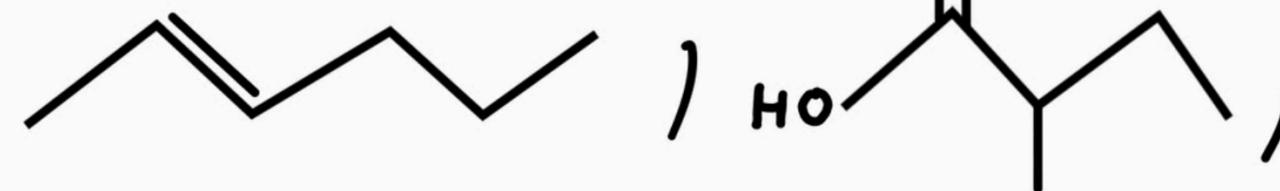


1º Identificação da Cadeia Principal

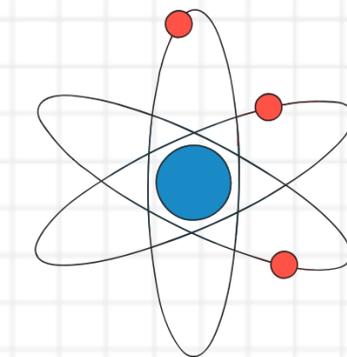
2º Numerar os Carbonos

Grupo Funcional > Insaturações > Ramificações

Ex:



# Regras para a Nomenclatura

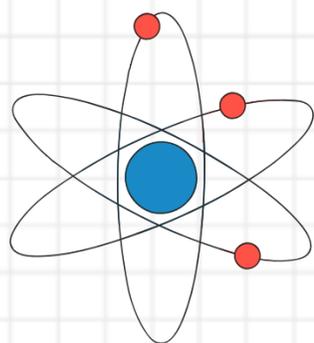
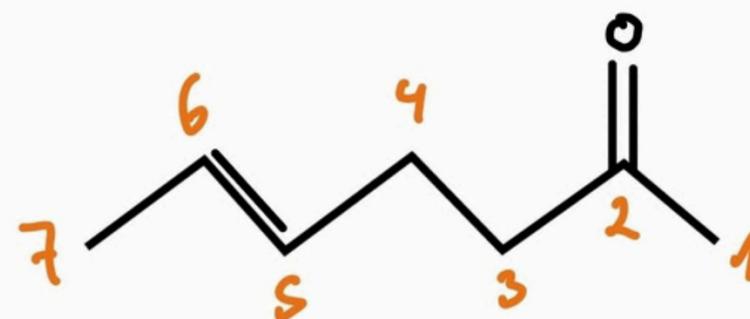
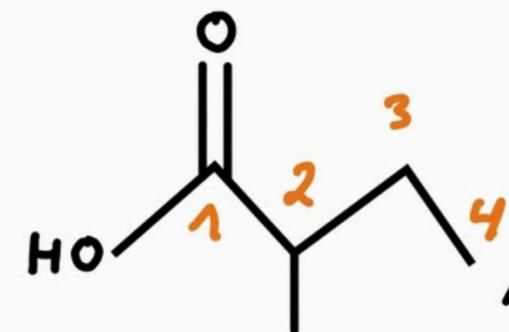
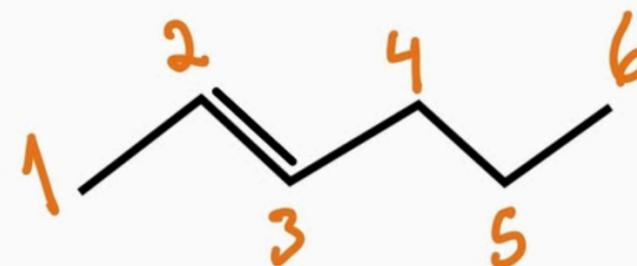


1º Identificação da Cadeia Principal

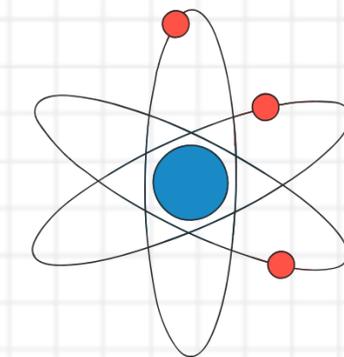
2º Numerar os Carbonos

Grupo Funcional > Insaturações > Ramificações

Ex:



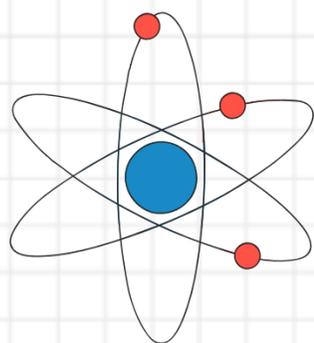
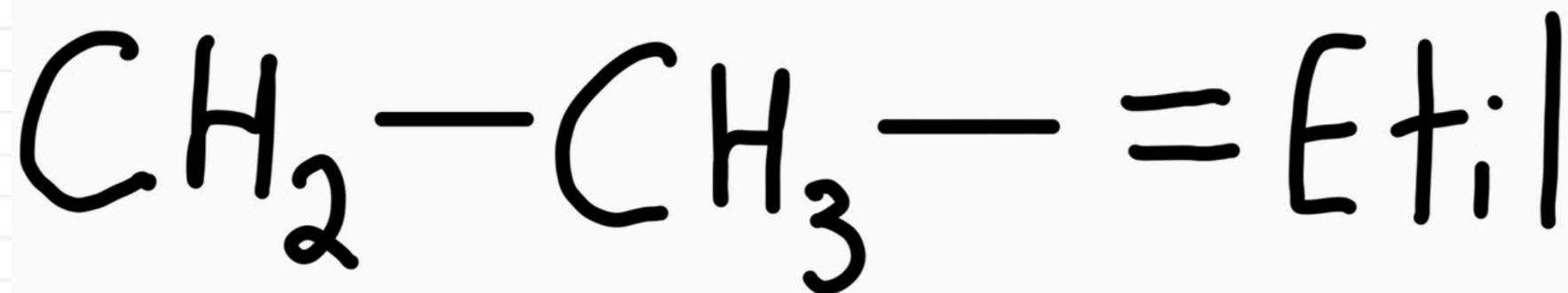
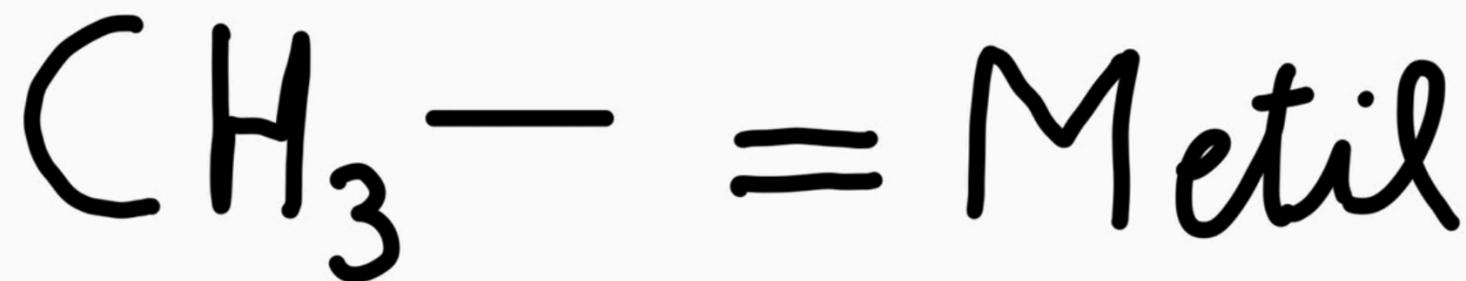
# Regras para a Nomenclatura



1º Identificação da Cadeia Principal

2º Numerar os Carbonos

3º Identificar, numerar e nomear os substituintes (Radicais)

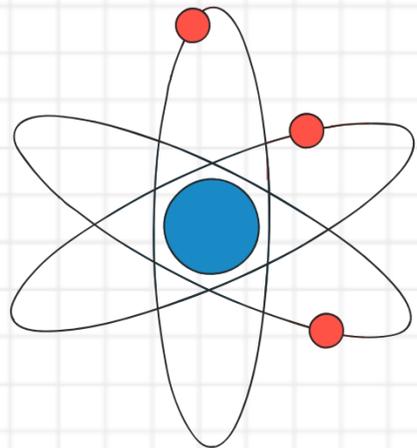
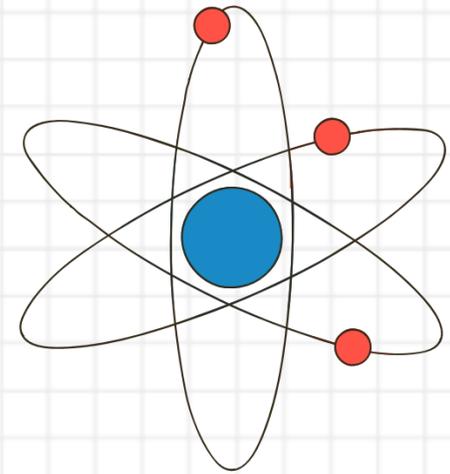


# Estrutura da Nomenclatura

Ramificações + Cadeia Principal

Ordem Alfabética

Prefixo + Sufixo + Infixo



# NOMENCLATURA

## Ramificação

METIL  
ETIL  
(EM ORDEM  
ALFABETICA)

## Prefixo

1C- MET  
2C- ET  
3C- PROP  
4C- BUT  
5C- PENT  
6C- HEX  
7C- HEPT  
8C- OCT  
9C- NON  
10C- DEC  
.....

## Sufixo

Lig. Simples- AN  
Lig. Dupla- EN  
Lig. Tripla- IN  
Duas duplas- DIEN

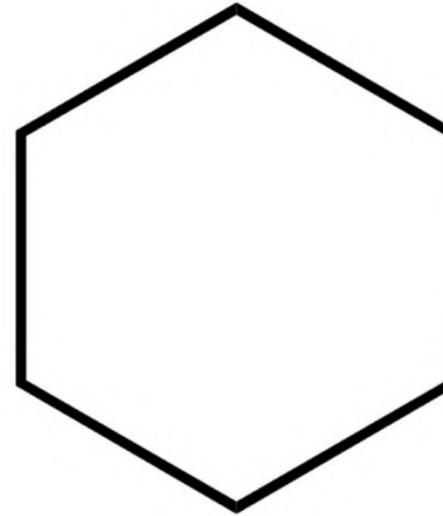
## Infixo

HIDROCARBONETO- O  
ÁLCOOL- OL  
FENOL- FENOL  
ALDEÍDO- AL  
CETONA- ONA  
ÁCIDO CARBOXÍLICO-  
OICO  
ÉTER- ÓXI + ANO  
ÉSTER- OATO + ILA  
AMINA- AMINA  
AMIDA- AMIDA

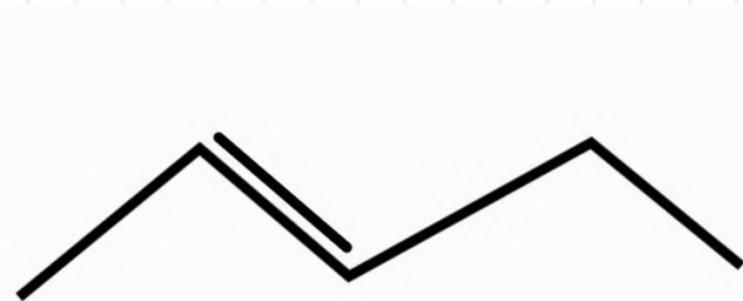
# Hidrocarbonetos



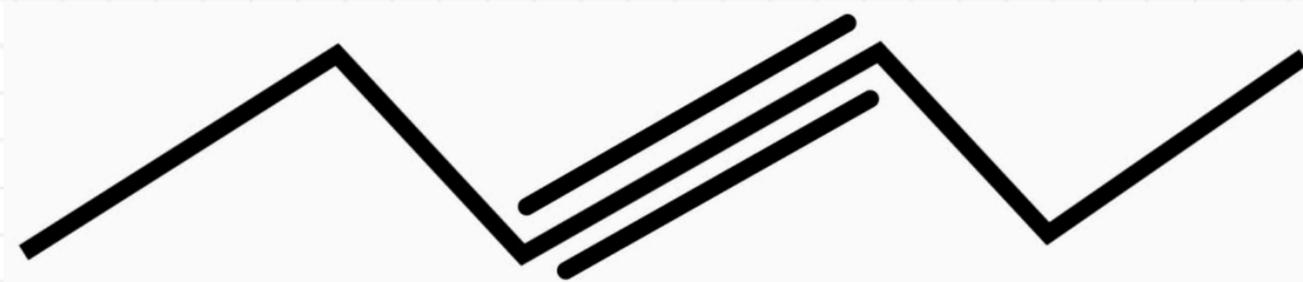
**BUTANO**



**Ciclo Hexano**

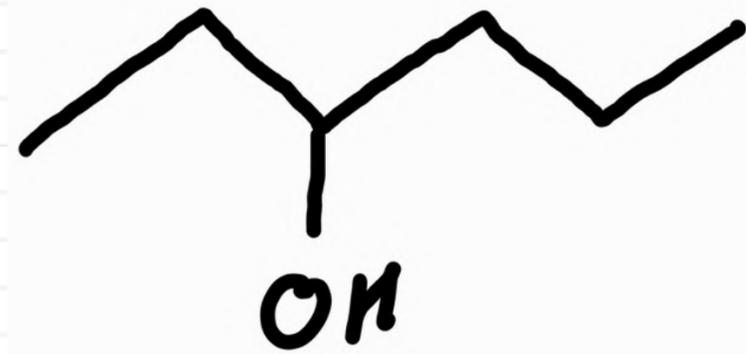


**PENT-2-ENO**

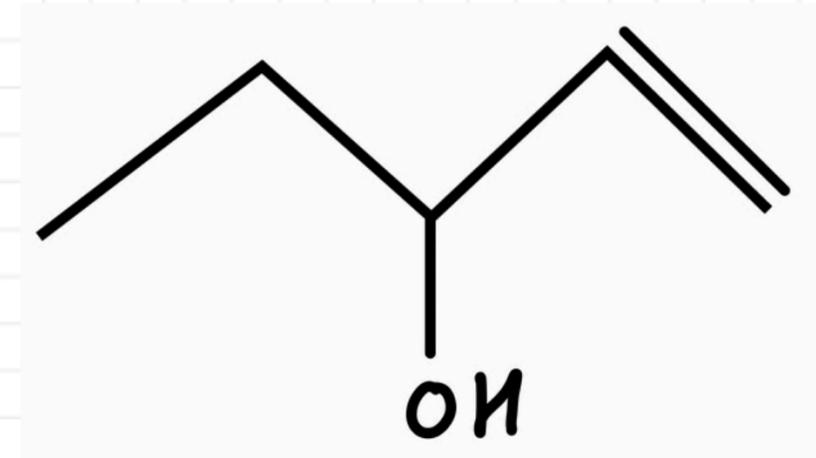


**HEX-3-INO**

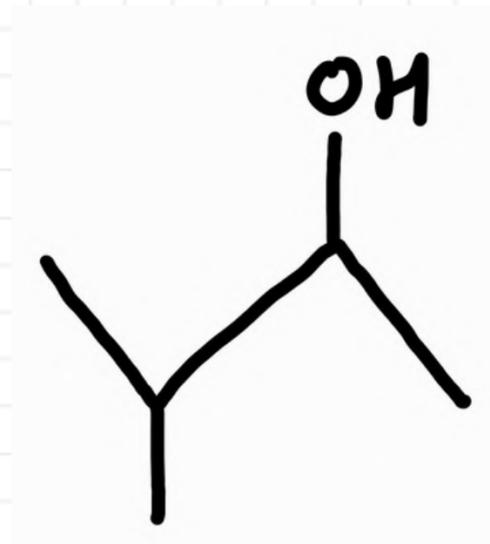
# Álcool



**HEXAN-3-OL**

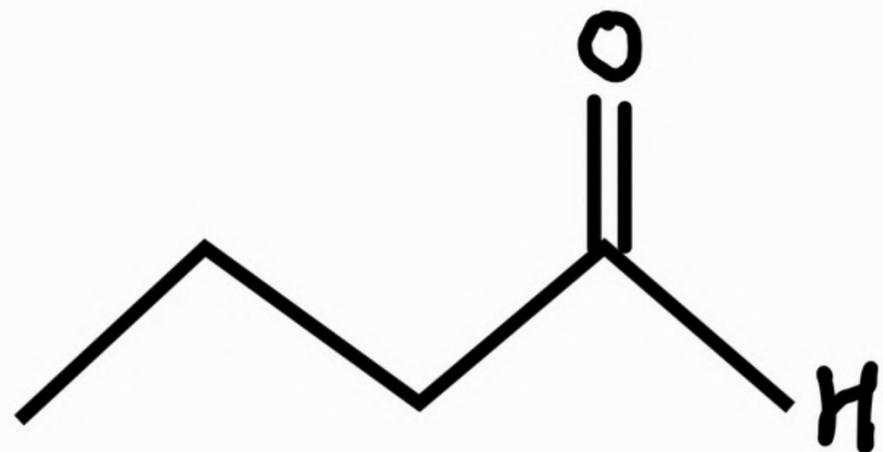


**PENT-4-EN-3-OL**

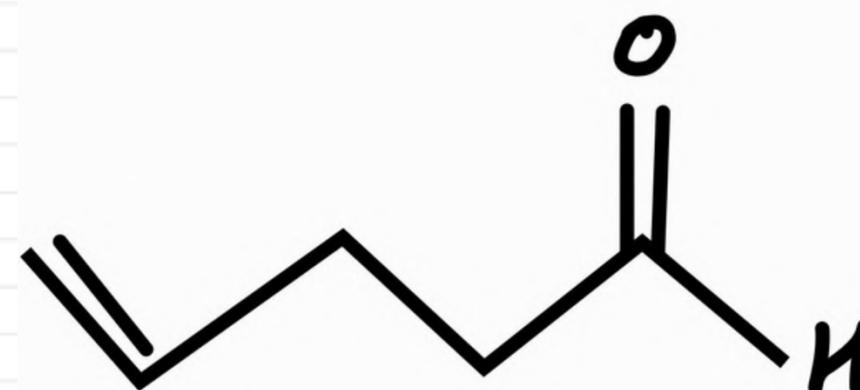


**4-METIL-BUTAN-2-OL**

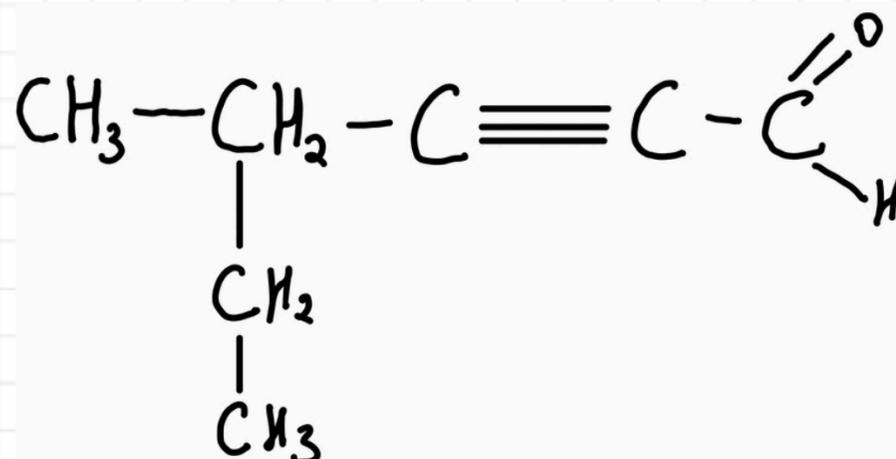
# Aldeído



**BUTANAL**

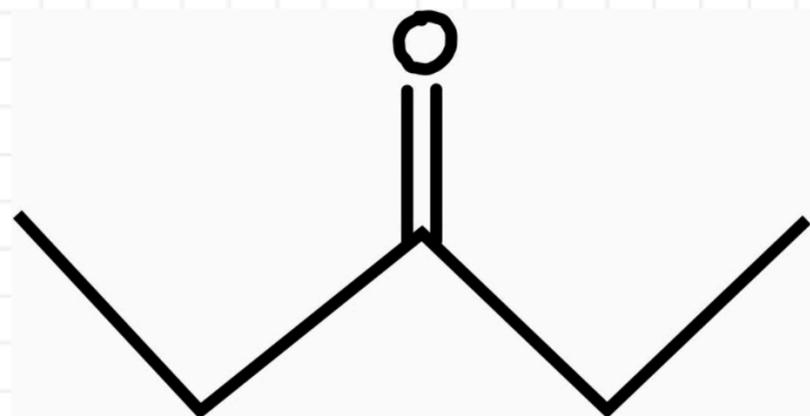


**PENT-4-ENAL**

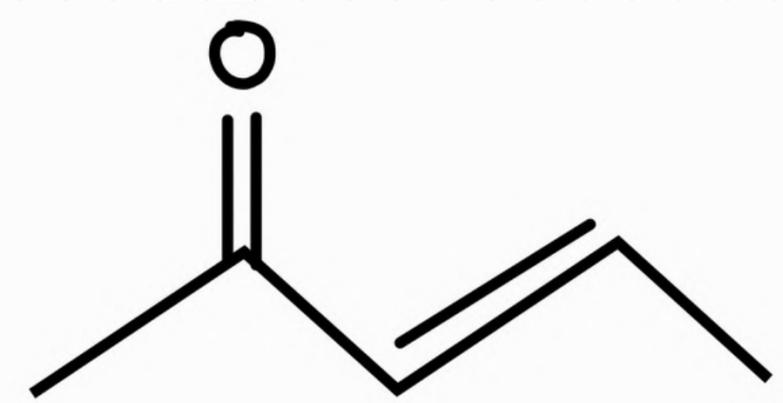


**4-METIL-HEX-2-INAL**

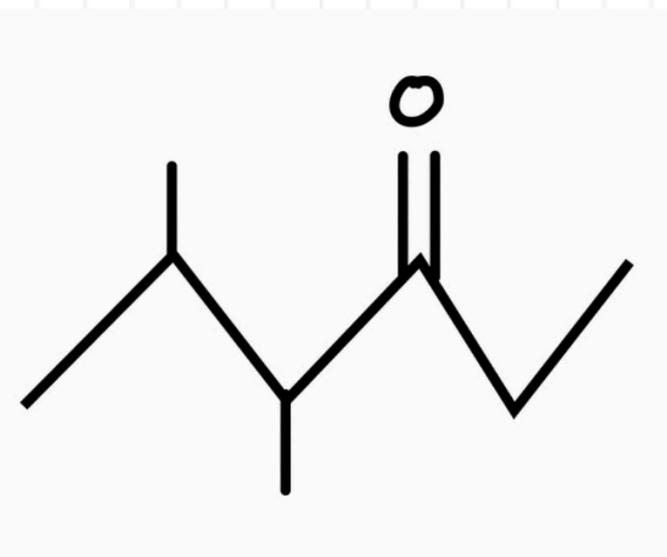
# Cetona



**PENTAN-3-ONA**

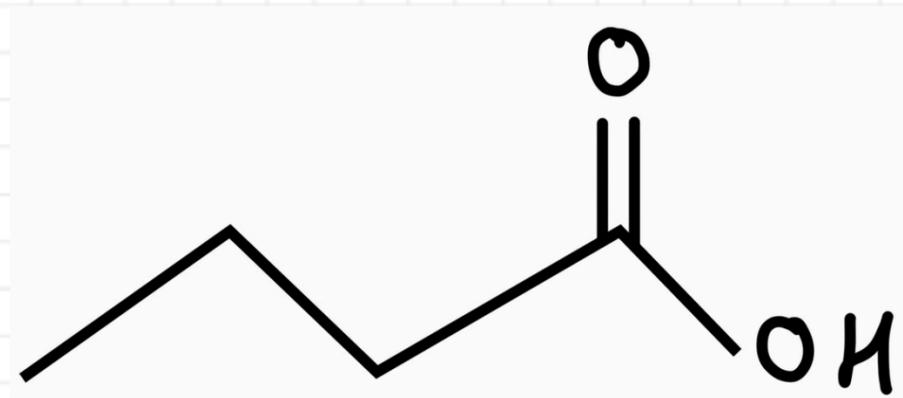


**PENT-3-EN-2-ONA**

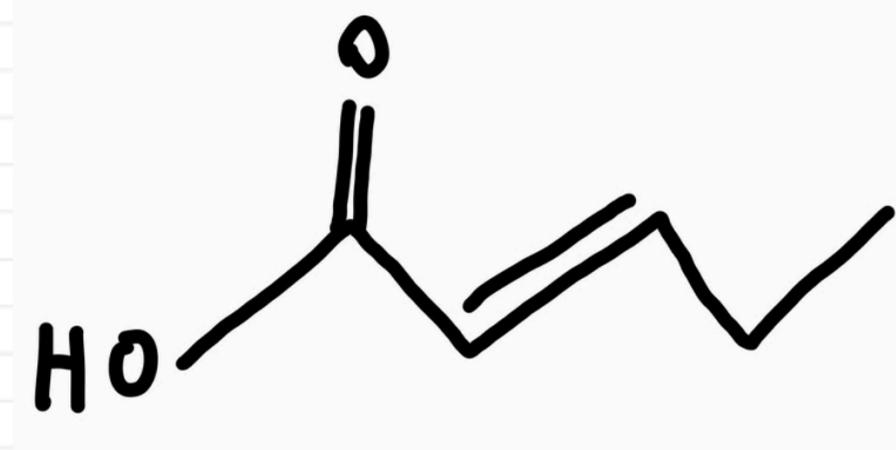


**3,4-METIL-HEXAN-3-ONA**

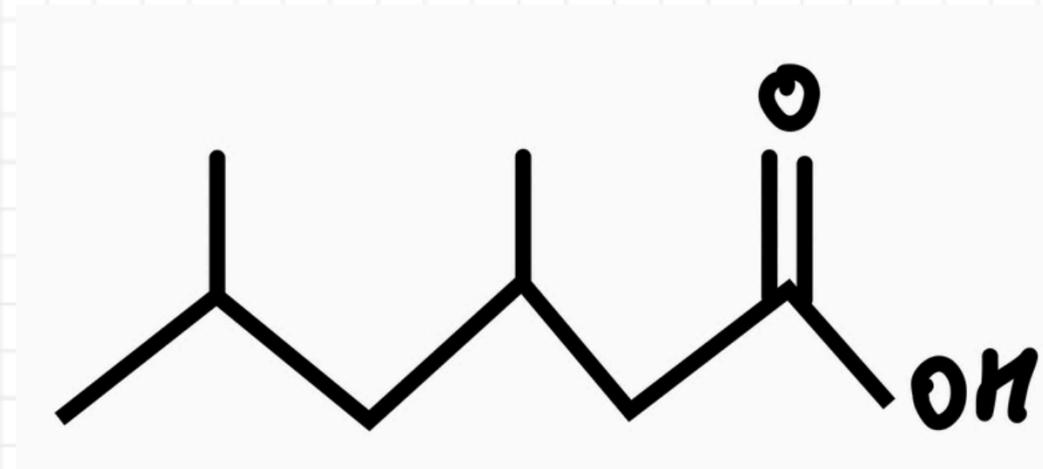
# Ácido Carboxílico



**ÁCIDO BUTANOICO**



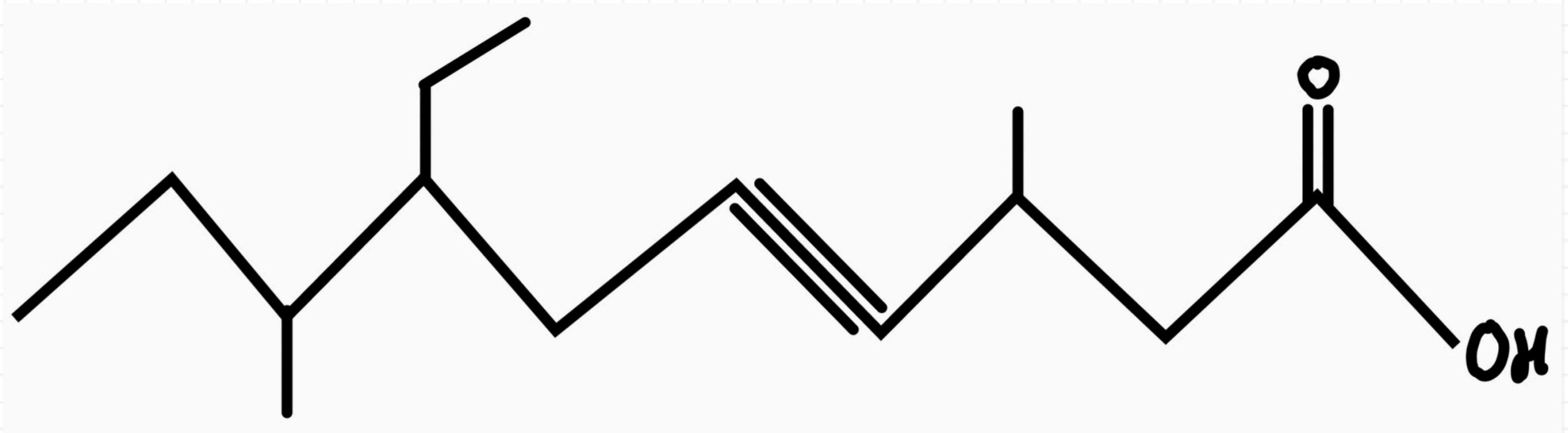
**ÁCIDO PENT-3-ENOICO**



**ÁCIDO 3,5-DIMETIL-HEXANOICO**

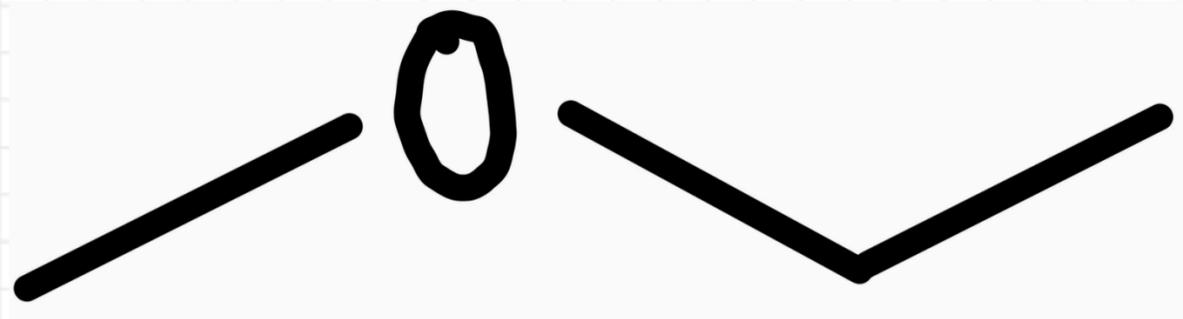
# Ácido Carboxílico

DESAFIO!!!!

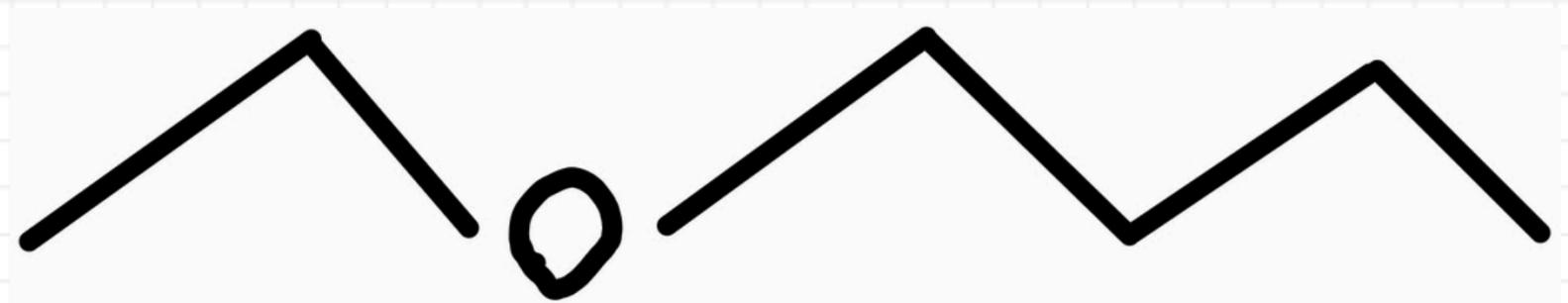


ÁCIDO 7-ETIL-3,8-DIMETIL-DEC-4-INOICO

# Éter

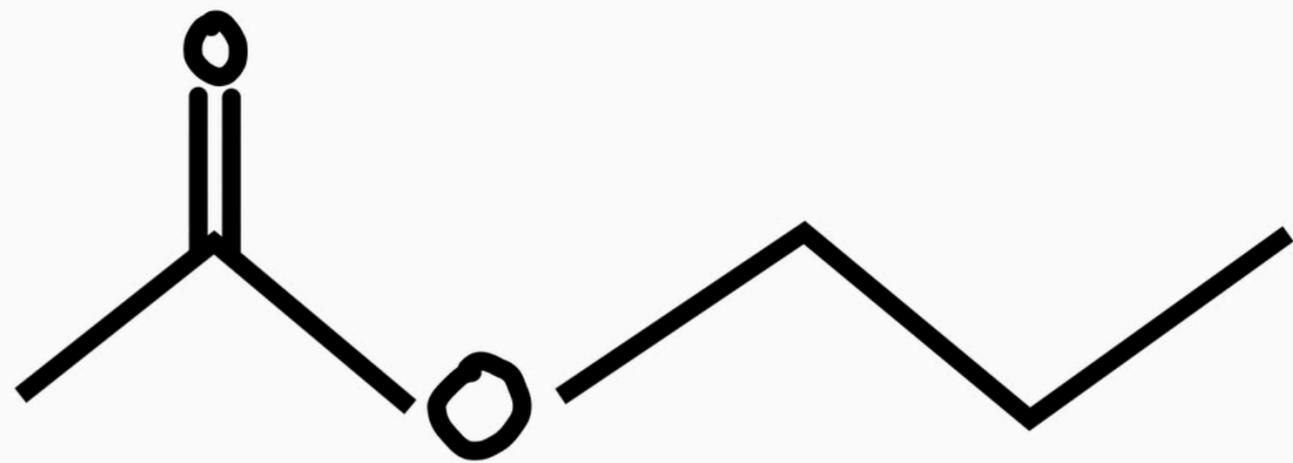


**METOXIETANO**

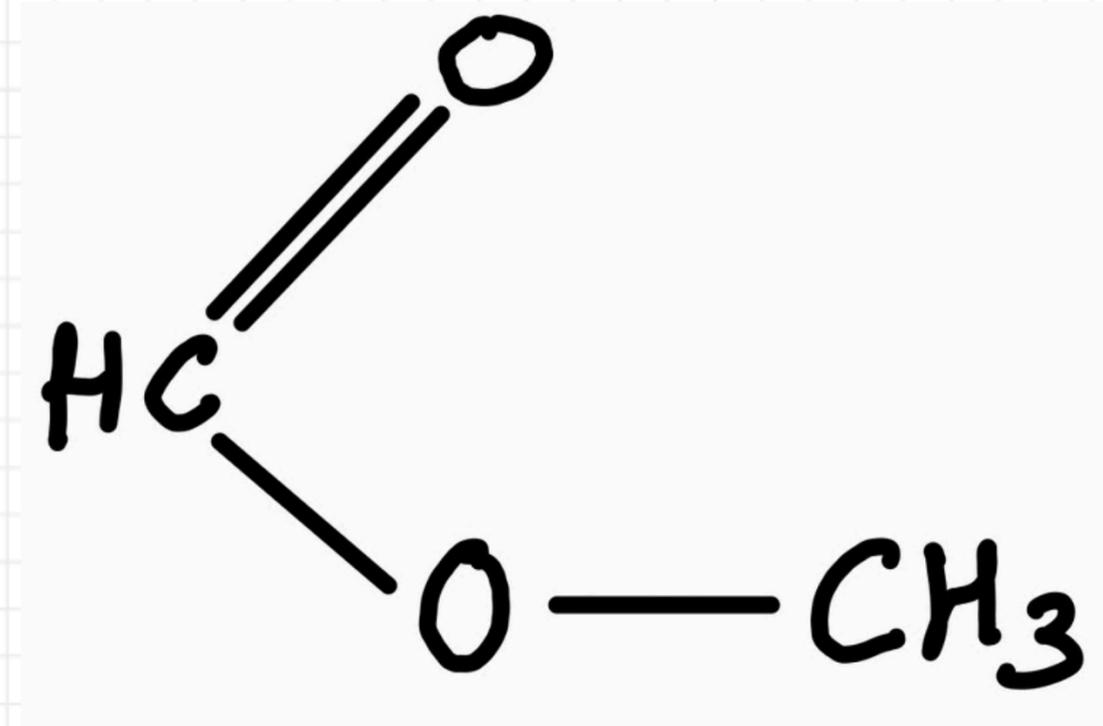


**ETOXIBUTANO**

# Éster

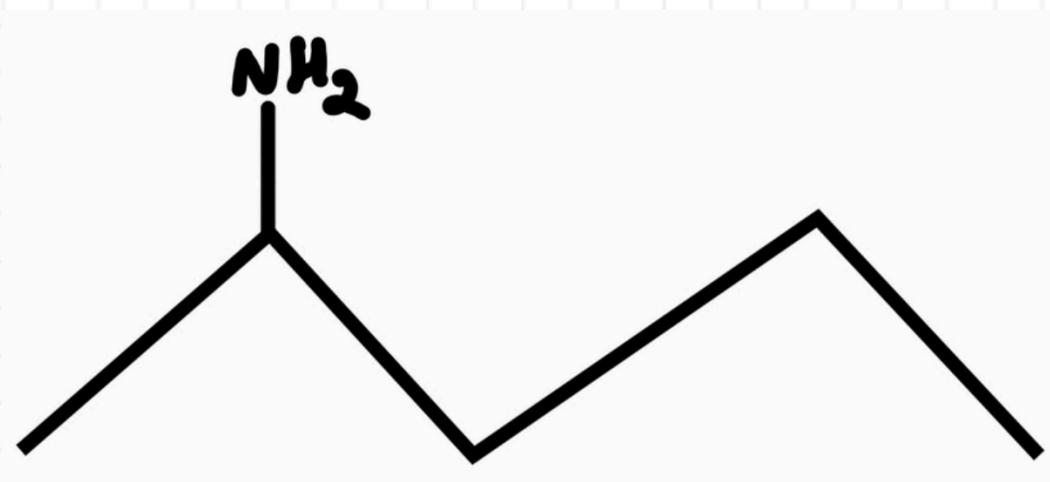


**PROPANOATO DE ETILA**

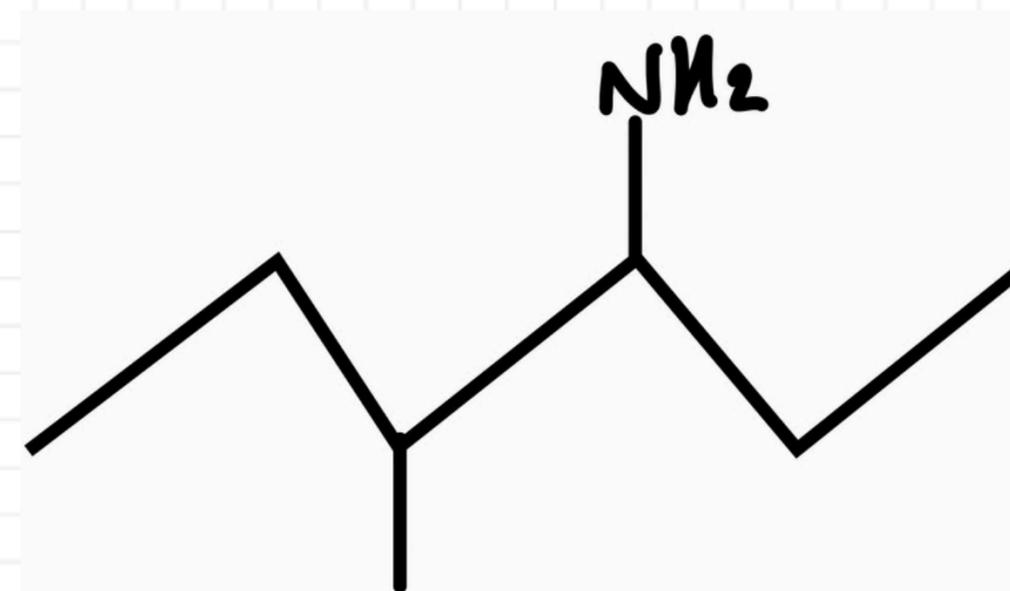


**METANOATO DE METILA**

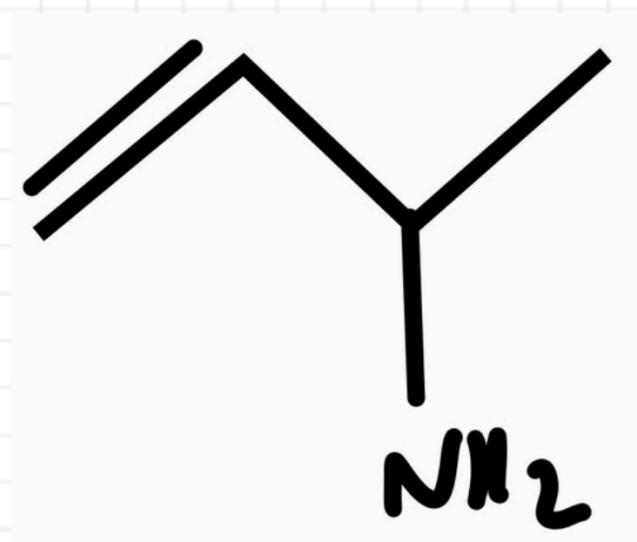
# Amina



**PENTAN-2-AMINA**

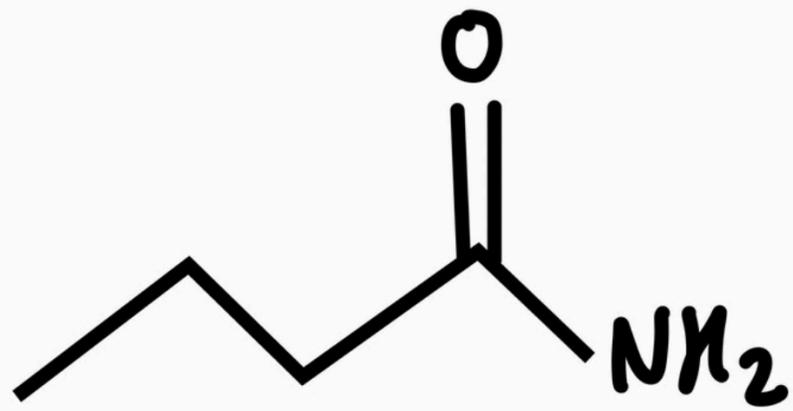


**4-METIL-HEXAN-3-AMINA**

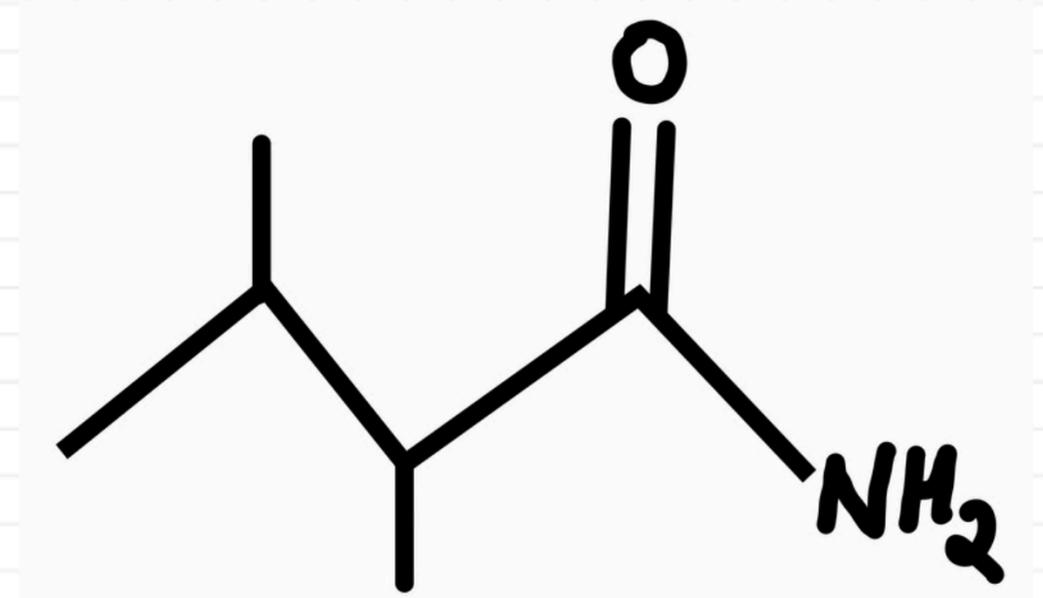


**BUT-3-EN-2-AMINA**

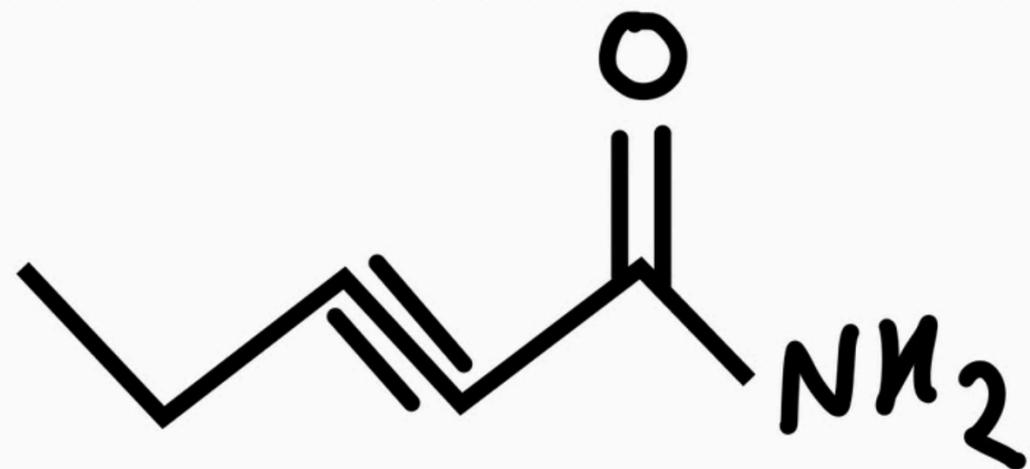
# Amida



**BUTANAMIDA**



**2,3-DIMETIL-BUTANAMIDA**



**PENT-2-INAMIDA**

# Exercícios

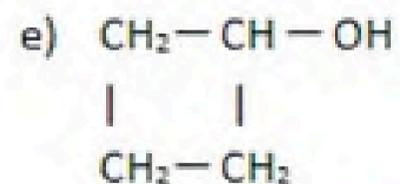
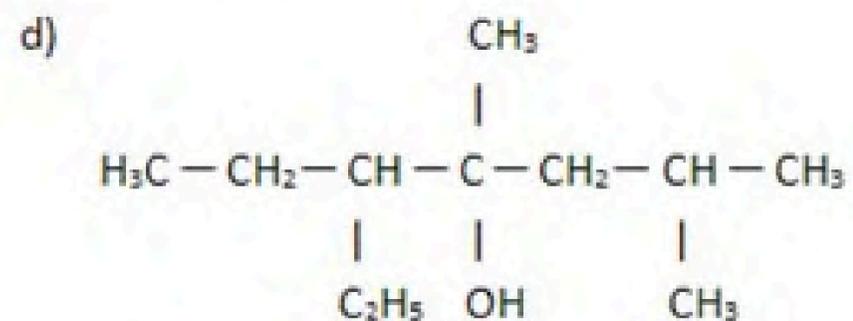
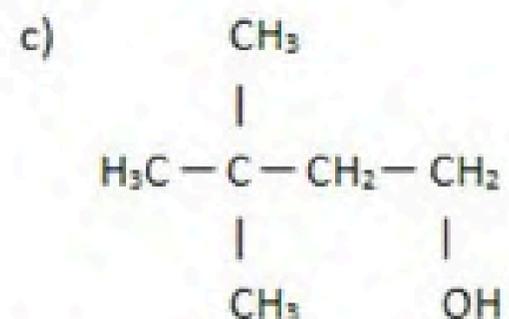
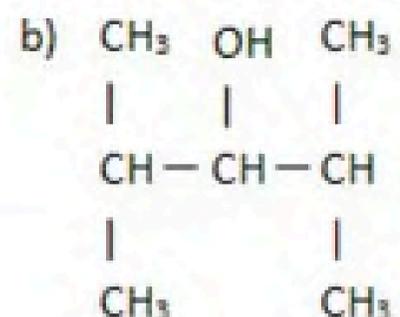
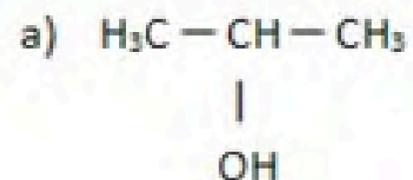
Na Química orgânica, os compostos são reconhecidos pelas cadeias formadas por carbono e hidrogênio. Entretanto, outros elementos podem fazer parte da estrutura química desses compostos, como o oxigênio.

Selecione a alternativa em que os dois compostos orgânicos apresentam funções orgânicas oxigenadas.

- a) clorofórmio e metanoato de etila
- b) propanol e ácido propanoico
- c) eteno e etanodiol
- d) etanamida e benzeno

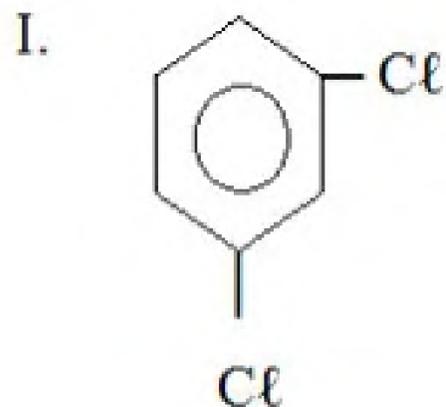
# Exercícios

Dê os nomes, segundo as regras estabelecidas pela IUPAC (União Internacional de Química Pura e Aplicada), dos seguintes compostos:



# Exercícios

(ENCE-UERJ-Cefet) Observe as fórmulas estruturais a seguir:



Segundo a ordem em que aparecem, sua nomenclatura oficial é:

- 1,3-diclorobenzeno, ácido etanoico, dietilamida, 1-buteno e propanoato de etila.
- ácido etanoico, 1,3-diclorobenzeno, dietilamina, 3-buteno e propenoato de etila.
- 2,4-diclorobenzeno, ácido etanoico, dietilamida, 2-buteno e etanoato de propenila.
- 1,3-diclorobenzeno, ácido etanoico, dietilamina, 1-buteno e propenoato de etila.
- 2,4-diclorobenzeno, ácido etanoico, dietilamina, 2-buteno e propanoato de etila.



Muito

Obrigado